Studies towards the synthesis of epothilone A *via* **organoboranes†**

P. Veeraraghavan Ramachandran,* J. Subash Chandra, Bodhuri Prabhudas, Debarshi Pratihar and M.Venkat Ram Reddy

Herbert C. Brown Center for Borane Research, Department of Chemistry, Purdue University, West Lafayette, IN, USA 47907-2084. E-mail: chandran@purdue.edu; Fax: 1 (765) 494 0239; Tel: 1 (765) 494 5303

Received 9th June 2005, Accepted 24th August 2005 First published as an Advance Article on the web 15th September 2005

Studies towards the synthesis of epothilone A *via* organoboranes have been described. A modified procedure for the large-scale preparation of $B-\gamma$, γ -dimethylallyldiisopinocampheylborane from prenyl alcohol has been developed. This reagent, upon reaction with various aldehydes, provides the corresponding α , α -dimethylhomoallylic alcohols in high enantioselectivities. The application of this reagent for the synthesis of the C_1-C_6 subunit of epothilone has been demonstrated. Alternatively, inter- and intramolecular asymmetric reduction protocols have also been utilized for the synthesis of the C_1-C_6 subunit of epothilone A. The synthesis of the C_7-C_{21} fragment of epothilone A involving asymmetric alkoxyallyl- and crotylboration using a-pinene-derived reagents has also been described.

Introduction

Microtubules play an important role in normal cellular processes and have become a key target for cancer chemotherapeutic drugs. The success of Paclitaxel (TaxolTM) as an anti-cancer drug has led to an invigorated quest for novel compounds with a mechanism of action similar to that of Taxol, with greater efficacy towards Taxol-resistant cells. One such compound is epothilone, which not only binds to the microtubules in a paclitaxel-like manner, but is much more active.**¹** Naturally occurring epothilone is a mixture of epothilone A and B (Fig. 1). Epothilone A (**1a**) turned out to be as active as Paclitaxel, whereas epothilone B is 50 times more active.^{1,2*a*-*c*} This simple structure and promising biological properties led to extensive efforts towards the synthesis of epothilone**²** and its analogs.**³**

Fig. 1 Naturally occurring epothilones.

As part of our program on the development and applications of pinane-based versatile organoborane reagents**⁴** for organic synthesis,**⁵** we undertook the synthesis of epothilones.**⁶** Our retrosynthetic analysis of **1a** is shown in Fig. 2.We envisaged that the required subunits $C_1 - C_6$ (2) and $C_7 - C_{21}$ (3) could be prepared *via* a-pinene-derived asymmetric alkoxyallyl- and crotylboration and dimethylallylboration protocols. We also envisioned that the total synthesis of epothilone could be achieved *via* an aldol coupling between the ethyl ketone **2** and the aldehyde **3**, followed by a macrolactonization under Yamaguchi conditions. Several approaches towards the synthesis of these two subunits are described in this manuscript. The C_1-C_6 subunit 2 was realized *via* an asymmetric dimethylallylboration strategy.Wittig olefination and alkoxyallyl- and crotylboration chemistry was used for the synthesis of **3**.

† Electronic supplementary information (ESI) available: ¹H and ¹³C NMR spectra. See http://dx.doi.org/10.1039/b508001k

Fig. 2 Retrosynthetic analysis for epothilone A.

Results and discussion

Synthesis of the C_1-C_6 **subunit of epothilone**

B-Chlorodiisopinocampheylborane is an excellent chiral reducing agent, which has been extensively used in the literature for the asymmetric reduction of a wide variety of ketones.**⁷** We visualized the application of $Ipc₂BCl$ for the intermolecular reduction of an acetylenic ketone**⁸** *en route* to the synthesis of the C_1-C_6 subunit of epothilone. We initiated the synthesis with the reaction of propionaldehyde and dimethylallylzinc bromide furnishing the homoallylic alcohol **10** (Scheme 1). Oxidation of the alcohol, followed by ozonolysis, afforded the keto-aldehyde **11** in 70% combined yield. Treatment with ethynylmagnesium bromide resulted in the chemoselective addition of the Grignard to the aldehyde, which upon oxidation yielded the diketone **12** in 80% yield.With our prior knowledge that the rate of reduction of tertiary ketones⁹ with Ipc₂BCl is extremely slow, the diketone 12 was reacted with $Ipc₂BCl$, and to our delight, reduction of the

Scheme 1 Chemoselective reduction of the acetylenic ketone. *Reagents* and conditions: (a) Zn, 80%; (b) DMP, 87%; (c) O₃, Me₂S, 80%; (d) HCëiCMgBr, 69%; (e) DMP, 80%; (f) Ipc₂BCl, 76% yield, 96% ee; (g) $Cy₂BH, CI₂BH, BH₃Me₂S, etc.$

acetylenic ketone occurred highly regioselectively (rather than reduction of the ethyl ketone), to give the propargylic alcohol **13** in 76% yield and 96% ee. At this stage, our efforts towards the hydroboration of alkyne did not materialize, and resulted in the formation of a complex mixture of products (Scheme 1). Having failed to achieve the hydroboration of the propargylic alcohol, we diverted our attention towards the intramolecular reduction of b-hydroxyketones for the synthesis of **2**.

Hydroxyketones**¹⁰** and keto-acids**¹¹** undergo intramolecular reduction with $Ipc₂BH$ (or $Ipc₂BCI$) yielding the corresponding diols or hydroxyacids respectively in very high ee. We planned the synthesis of the C_1-C_6 subunit *via* this intramolecular reduction protocol. Monosilylation of 1,3-propanediol, followed by Dess–Martin periodinane (DMP)**¹²** oxidation provided 3 silyloxypropionaldehyde **15** (Scheme 2). Allylation with the prenylzinc reagent furnished the homoallylic alcohol **16** in 71% yield. Oxidation and deprotection of the silyl ether afforded the b-hydroxyketone **17**. Intramolecular reduction of the bhydroxyketone with Ipc₂BH provided the diol 18 in 75% yield and in very high ee (98%). Sequential protection of the primary alcohol as the PMB ether and the secondary alcohol as the TBS ether yielded **19**. Ozonolytic cleavage of the olefin afforded the aldehyde, which upon reaction with ethylmagnesium bromide, followed by DMP oxidation yielded 2, the C_1-C_6 subunit of epothilone A (Scheme 2) in 77% yield.

Scheme 2 Intramolecular reduction of the b-hydroxy ketone. *Reagents and conditions*: (a) TBSCl, imidazole, 89%; (b) DMP, 81%; (c) Zn, 71%; (d) DMP, 83%; (e) HCl, 80%; (f) (+)-Ipc₂BH, NaOH, H₂O₂, 75% yield, 98% ee; (g) $\text{MeOC}_6\text{H}_4\text{CH}_2\text{OC}(\text{NH})\text{CCl}_3$, CSA, 80% ; (h) TBSCl, imidazole, 86% ; (i) O₃, Me₂S, 78%; (i) EtMgBr, 76%; (k) DMP, 77%.

Even though we were able to synthesize the subunit with high stereoselectivity, we were not satisfied with the overall yield and number of steps required for the synthesis. With the intent of overcoming these shortcomings, we concentrated our efforts on the development of an asymmetric dimethylallylboration protocol.

Improved procedure for the synthesis of *B***-** γ **,** γ **-dimethylallyldiisopinocampheylborane⁶***^a* **.** "Allyl"boration is one of the most important carbon–carbon bond forming reactions in organic synthesis.¹³ Over the past two decades, we have developed several highly functionalized "allyl"boranes based on terpenes.**⁴** While some of these reagents have been well utilized by the synthetic organic community for the total synthesis of a variety of complex natural products,**¹⁴** others have been sparsely used, probably due to the difficulty in their preparation or the cost of the starting materials. One such reagent is $B-\gamma, \gamma$ -dimethylallyldiisopinocampheylborane 20 ,¹⁵ which upon reaction with aldehydes provides α , α -dimethylhomoallylic alcohols. Several natural products, such as artemesia alcohol,**¹⁶** bryostatin,**¹⁷** epothilone, pederin,**¹⁸** and mycalamide**¹⁹** contain an α , α -dimethylhydroxy unit in them, and one could envisage the use of 20 for the preparation of the α , α -dimethylalcohol moiety. Following Brown's initial report of the synthesis of artemesia alcohol,**¹⁵** only Schinzer has described the application of **20** for the synthesis of epothilones.**²⁰** Brown's original preparation of **20** involved the hydroboration of relatively expensive 1,1 dimethylallene **21** with diisopinocampheylborane, thus making it impractical for large-scale applications (Scheme 3). The wide range of potential applications as well as the highly expensive starting materials involved in the preparation of **20** persuaded us to develop an inexpensive method for the large-scale synthesis of this reagent.

Scheme 3 Literature synthesis of **20**.

The traditional preparation of *B*-allyldialkylboranes involves the reaction of the corresponding allyl Grignard reagent with *B*halodialkylborane or *B*-methoxydialkylborane. We envisioned the use of dimethylallyl Grignard, which can be readily prepared from the commercially available and relatively inexpensive prenyl alcohol. The reaction of prenyl alcohol with PBr₃ produced the corresponding bromide. However, the reaction of the dimethylallyl bromide with magnesium turnings was not satisfactory and resulted in the predominant formation of homocoupling products. The subsequent addition of (+)- $Ipc₂BOMe$ to the reaction mixture resulted in very low yields of the allylborane **20** (∼10% on the basis of 11B NMR spectroscopy). Consequently, the bromide was replaced with chloride on the basis that the chloride might reduce the risk of homocoupling. Accordingly, the dimethylallyl chloride **22** was prepared by the reaction of prenyl alcohol with thionyl chloride. Temperature played a crucial role in the formation of the Grignard. When the temperature was too high, homocoupling dominated, and when the temperature was kept below 0 *◦*C, the formation of the Grignard reagent was arrested. Grignard formation was finally achieved by carefully keeping the reaction temperature in the range 0–5 *◦*C. Upon successful formation of the Grignard, the reaction mixture was transferred to a solution of $(+)$ -Ipc₂BOMe in ether at 0 \degree C to furnish the required dimethylallylborane **20** in essentially quantitative yield (Scheme 4).**⁶***^a* The methoxymagnesium chloride salt was filtered using a Kramer's filter under nitrogen, and the solvent was evaporated under vacuum to provide 90% yield of **20** (11B NMR peak at δ 79). Similarly, starting from (−)-Ipc₂BOMe, the antipode of the reagent **20** was prepared. The reagent was tested for its large-scale applicability by carrying out the reaction on a 0.5 mole scale – highly reproducible results were obtained.

Scheme 4 Modified procedure for the synthesis of **20**.

We then examined **20** for the allylboration of various aldehydes. The reaction with aliphatic aldehydes such as propionaldehyde **23a**, isobutyraldehyde **23b**, pivalaldehyde **23c** (Table 1, entries 1–3), took place smoothly and the product homoallylic **Table 1** Dimethylallylboration of aldehydes

alcohols **24a–c** were obtained in very good yields (81–88%) and in excellent ee (95–97%). The ee was determined by making the *p*-nitrobenzoate esters of the homoallylic alcohols, and analyzing them by HPLC. Similarly, benzaldehyde **23d** and cinnamaldehyde **23e** (entries 4 and 5) provided alcohols **24d** and **24e** in excellent yield and ee. The reaction with a-chiral ($23g$) and β -chiral aldehydes (entries 7 and 8) took place in a reagent-controlled manner, and the homoallylic alcohols **24g** and **24h** were obtained in 92% and 95% de, respectively.

The modified synthesis of dimethylallylborane was then utilized in the synthesis of the C_1-C_6 subunit 2 of epothilone. Dimethylallylboration of aldehyde **23f** using **20** at −100 *◦*C furnished the homoallylic alcohol **24f** in 82% yield and 95% ee. The completion of the synthesis of the C_1-C_6 subunit was achieved without any difficulty by a silyl protection, ozonolysis, alkylation, and oxidation sequence as described earlier (Scheme 2).

Synthesis of the C_7-C_{21} **subunit of epothilone**

We envisaged the synthesis of the C_7-C_{21} subunit (3) by the convergence of the three subunits C_7-C_{12} (25), $C_{13}-C_{16}$ (26), and $C_{17}-C_{21}$ (**5**) using a sequential Wittig coupling protocol. Our initial plan was to couple thiazolyl chloride **5** with **26** to yield **27**, followed by a second Wittig coupling with the primary iodide **25** (path A, Fig. 3). Alternatively, a Wittig coupling between **25** and **26** would provide **28**, which after another Wittig coupling with thiazolyl chloride should also provide the required C_7-C_{21} subunit of epothilone (path B, Fig. 3). Our results are discussed below.

The C_7-C_{12} segment (iodide 25) was prepared as shown in Scheme 5. Silyl protection of the commercially available (*R*)-3 hydroxy-2-methylpropionate (Roche ester, **29**), reduction of the ester using BH_3 ·Me₂S, and iodination with I_2/PPh_3 furnished the iodide **30**. Nucleophilic substitution of the iodide with higherorder allyl cuprate,**²¹** generated by the reaction of allyl Grignard with $Li₂CuCl₄$ or CuI, provided the olefin 31 in 22–50% yield. Attempts towards obtaining a higher yield for the substitution proved futile even after an extensive study of reagents and conditions. Hydroboration-oxidation of the olefin **31** to the primary alcohol, followed by iodination provided the required iodide **25** (Scheme 5).

Scheme 5 Preparation of the C_7-C_{12} segment 33. *Reagents and conditions*: (a) TBSCl, imidazole, DMF, $0 °C$, 90%; (b) BH₃·Me₂S, NEt₃, MeOH, THF, 25 °C, 78%; (c) I₂, PPh₃, imidazole, CH₂Cl₂, 8 h, 87%; (d) allylMgBr, Li₂CuCl₄, THF, −78 °C, 30–50%; (e) BH₃·Me₂S, 3 h, NaOH, H_2O_2 , 3 h, 40%; (f) I_2 , PPh₃, imidazole, CH₂Cl₂, 8 h, 80%.

The synthesis of $C_{13}-C_{16}$ segment proved to be challenging. The synthesis was initiated with the alkoxyallylboration**²²** of acetaldehyde with $B-\gamma$ -methoxyethoxymethoxyallyldiisopinocampheylborane to furnish the homoallylic alcohol **33** (Scheme 6). The corresponding alkoxyallylborane reagent was prepared by the reaction of lithiated allyl methoxyethoxymethyl ether **32** with *B*-methoxydiisopinocampheylborane. Attempts towards the oxidation of the homoallylic alcohol under Jones $(CrO₃)$ or Swern $[Me₂SO, (COCl)₂]$ conditions, to our dismay, led to isomerization vielding the α , β -unsaturated ketone 34. Oxidation of the homoallylic alcohol **33** was finally achieved with Dess–Martin periodinane to yield the ketone **35** in 76% yield. Hydroboration of **35** proved futile and we could not

Fig. 3 Retrosynthetic analysis for the C_7-C_{21} subunit.

Scheme 6 Preparation of the C₁₃–C₁₆ segment 38. *Reagents and conditions*: (a) *s*-BuLi, THF, 0.5 h, (+)-Ipc₂BOMe, 1 h, BF₃·Et₂O, 5 min, CH₂CHO, -78 °C, 3 h, (ii) NaOH, H₂O, 3 h, 75% vield, 95% ee: ^{−78} °C, 3 h, (ii) NaOH, H₂O₂, 3 h, 75% yield, 95% ee; (b) CrO3, CH2Cl2; (c) Me2SO, (COCl)2, NEt3; (d) DMP, CH2Cl2, 25 *◦*C, 76%; (e) BH₃·THF, Cy₂BH, 9-BBN, Cl₂BH, CatBH/RhCl(PPh₃)₃ *etc.* (f) Cy2BH, NaOH, H2O2, 76%; (g) (i) Br2, Na2CO3, MeOH, 0 *◦*C; or (ii) Br₂, HMPT, NaHCO₃, H₂O, CH₂Cl₂, 25 °C; or (iii) (Bu₃Sn)₂O, Br₂, CH₂Cl₂, 25 °C; or (iv) NaOCl, CH₃COOH, 25 °C; or (v) KBrO₃, N_4 HSO₃, CH₃CN, H₂O; or (vi) (NH_4) ₂Ce (NO_3) ₆, KBrO₃, CH₃CN, 80 [°]C *etc.* (h) TBSCl, imidazole, DMF, 25 [°]C, 87%; (i) DMP, CH₂Cl₂, 25 °C, 80%.

realize the formation of the hydroxyketone **36**, irrespective of the borane reagents used, such as BH_3 , Cy_2BH , $9-BBN$, Cl_2BH , catecholborane *etc.* (Scheme 6).

Selective oxidation of secondary alcohols in the presence of a primary alcohol is a well-studied field of chemistry,**²³** and we attempted to use this reaction in our synthesis. Hydroboration of the homoallylic alcohol **33** was achieved by treatment with 2.5 eq. of dicyclohexylborane, and the 1,4 diol **37** was obtained in 76% yield and >98% regioselectivity. However, selective oxidation of the secondary alcohol in **37** did not work under a variety of conditions. Oxidizing agents such as Br_2/Na_2CO_3 , $Br_2/HMPT$, $Br_2/(Bu_3Sn)_2O$, NaOCl, KBrO3/NaHSO3, CAN/KBrO3 *etc.* did not provide the required hydroxyketone. As a final resort, a two-step indirect formation of the $C_{13}-C_{16}$ subunit 38 was achieved by selective protection of the primary alcohol as the silyl ether, followed by oxidation of the secondary alcohol to the corresponding ketone **38** in 70% combined yield (Scheme 6).

Having achieved the synthesis of the segments C_7-C_{12} (25, Scheme 5), $C_{13}-C_{16}$ (38, Scheme 6), and $C_{17}-C_{21}$ (5²⁴), we proceeded further towards the coupling of these pieces by path A, Fig. 3. Formation of the Wittig salt was realized by refluxing **5** with tri-*n*-butylphosphine (the Schlosser modification of the Armstrong protocol).**²⁵** Treatment of the Wittig salt with NaHMDS, followed by the addition of the ketone **38** furnished the olefin **39**. Protolytic cleavage of the silyl ether, followed by DMP oxidation provided the aldehyde **27**. Wittig coupling of the aldehyde 27 with the $C_7 - C_{12}$ subunit (iodide, 25) under standard highly basic conditions led to extensive β -elimination of the alkoxy (MEM) group to furnish the conjugated trienyl thiazole **42** as the major coupling product (Scheme 7).

With the intent of increasing the yield for the Wittig coupling, we used the alternative Wittig olefination protocol shown in path B, Fig. 3. Partial oxidation of the primary alcohol in the diol **37** yielded the γ -lactol, **43**. Treatment of this hemiacetal with the Wittig ylide generated from the primary iodide **25** resulted in very low yields of the olefinic alcohol **44** (Scheme 8). Having failed to optimize the yields in the Wittig olefination of the γ lactol, we resorted to a protected aldehyde for the coupling. Thus, the homoallylic alcohol **33** was protected as the silyl ether **45**. Hydroboration of the olefin **45** with dicyclohexylborane furnished the primary alcohol in >98% regioselectivity. Oxidation of the primary alcohol to the aldehyde **46**, followed by a Wittig coupling with the ylide obtained from the primary iodide **25**, yielded the olefin **47** in \lt 5% yield (Scheme 8).

In order to overcome the low yields in the Wittig coupling in our synthesis, we used the relatively non-labile benzyl ether as the protecting group. Thus, the protection of the homoallylic alcohol **33** as its benzyl ether, followed by hydroboration and oxidation using DMP yielded aldehyde **7** (Scheme 9). Although one of the chiral centers derived from alkoxyallylboration will be converted

Scheme 7 Coupling of **5**, **25** and **38**. *Reagents and conditions*: (a) (i) PBu3; (ii) NaHMDS, 74%; (b) AcOH, 83%; (c) DMP, 85%; (d) (i) **25**, PPh3; (ii) NaHMDS.

Scheme 8 Wittig olefination of γ -lactol 55 with iodide 33. *Reagents and conditions*: (a) (i) Cy₂BH, NaOH, H₂O₂, 76%; (ii) TPAP, NMO, 85%; (b) (i) 25, PPh3; (ii) NaHMDS; (c) TBSCl, imidazole, 25 *◦*C , 89%; (d) Cy2BH, NaOH, H2O2, 25 *◦*C, 3 h, 78%; (e) DMP, 25 *◦*C, 0.3 h, 97%.

Scheme 9 Preparation of the $C_7 - C_{21}$ subunit. *Reagents and conditions*: (a) NaH, BnBr, THF, 0–25 *◦*C , 78%; (b) BH3·Me2S, 0 *◦*C, 4 h, NaOH, H₂O₂, 25 °C, 3 h, 82%; (c) DMP, CH₂Cl₂, 25 °C, 0.3 h, 97%; (d) (i) **25**, PPh3, benzene, 80 *◦*C, 12 h; (ii) NaHMDS, HMPA, THF, −78 *◦*C, (iii) **7**, 25 *◦*C, 58%; (e) Li, NH3(l), THF, −78 *◦*C, 3.5 h, MeOH–NH₄Cl, 25 °C, 15 h, 85%; (f) DMP, CH₂Cl₂, 25 °C, 8 h, 95%; (g) (i) **5**, PBu3, 3 h, 70 *◦*C; (ii) NaHMDS, THF, 0 *◦*C, 0.8 h; (iii) **28**, 0 *◦*C, 6 h, 98%; (h) AcOH–H2O–THF (3 : 1 : 1), 25 *◦*C, 18 h, 80%; (i) DMP, CH₂Cl₂, 25 ^{*ό*}C, 3 h, 98%.

to the ketone at a later stage of the synthesis, the protocol does not necessitate any additional steps to create this chiral center. Wittig coupling of aldehyde **7** with the primary iodide **25** provided the olefin **48** in 58% yield (∼9:1 *Z* : *E*). Debenzylation, under Birch reduction conditions, furnished the alcohol, which upon DMP oxidation provided the ketone **28** in 95% yield.

A modified Wittig coupling between thiazolyl chloride **5** and the ketone **28** yielded **49**. Silyl deprotection and DMP oxidation furnished the required $C_7 - C_{21}$ subunit **3** of epothilone in 98% yield.

Even though the synthesis of the C_7-C_{21} subunit of epothilone was achieved, the unreliable yields during the allyl substitution of the primary iodide **30** compelled us to modify the synthetic scheme. We designed a protocol utilizing α pinene-based crotylboration.**²⁶** This procedure is advantageous because it replaces the relatively expensive Roche ester **29**. We began our synthesis with the *Z*-crotylboration of *tert*butyldimethylsiloxyacetaldehyde **50** to furnish the homoallylic alcohol **51**. Silyl protection, followed by oxidative cleavage of the olefin furnished the aldehyde **52**. Wittig coupling of **52** with 3-benzyloxypropyl iodide,**²⁷** under standard conditions, led to the stereoselective formation of *Z*-olefin **53**. Pd-catalyzed hydrogenation resulted in the simultaneous debenzylation and reduction of the double bond yielding the primary alcohol, which upon iodination provided **6**. Formation of the Wittig ylide of **6**, followed by treatment with the aldehyde **7**, furnished the coupled olefin **54**. It is noteworthy that the *Z*-olefin was formed selectively and in high yield (85%) from the substrate, which is prone to B-elimination, as was the case in the earlier protocols. Birch reduction afforded the debenzylated secondary alcohol, which underwent oxidation, yielding the ketone **4**. Wittig coupling of the ketone **4** with thiazolyl chloride **5** using the Schlosser modification of the Armstrong protocol provided **55** in 83% yield. Selective deprotection of the two silyl groups using dilute HCl, followed by $Pb(OAc)_4$ cleavage provided the required $C_7 - C_{21}$ subunit **3** (Scheme 10).

Scheme 10 Revised procedure for the preparation of the $C_7 - C_{21}$ subunit. *Reagents and conditions*: (a) (i) *n*-BuLi, *cis*-2-butene, KO*^t* Bu, THF, −45 °C, 0.3 h; (+)-Ipc₂BOMe, −78 °C, 1 h; BF₃·Et₂O, 5 min; **67**, −78 *◦*C, 3 h; (ii) NaOH, H2O2, 25 *◦*C, 3 h, 82%; (b) TBSCl, imidazole, DMF, 25 *◦*C, 9 h, 89%; (c) (i) OsO4, NMO, acetone–water (3 : 1), 25 *◦*C, 4 h; (ii) NaIO4, acetone–water (3 : 1), 25 *◦*C, 2.5 h, 90%; (d) (i) **71**, PPh3, *p*-xylene, 150 *◦*C, 12 h; (ii) NaHMDS, 0 *◦*C, THF, **70**, 3 h, 77%; (e) H_2 , Pd/C, EtOAc, 8 h, 60%; (f) I₂, PPh₃, imidazole, CH₂Cl₂, 8 h, 78%; (g) (i) **6**, PPh3, *p*-xylene, 150 *◦*C, 3 h; (ii) NaHMDS, HMPA, THF, −78 *◦*C, 2 h; (iii) 7, −78–25 *◦*C, 15 h, 85%; (h) Li, NH3(l), THF, −78 *◦*C, 3.5 h, MeOH–NH₄Cl, 25 °C, 15 h, 95%; (i) DMP, CH₂Cl₂, 25 °C, 0.3 h, 98%; (j) (i) **5**, PBu₃, 3 h, 70 °C; (ii) NaHMDS, THF, 0 °C, 0.7 h; (iii) **4**, 0 *◦*C, 15 h, 83%; (k) THF–H2O–HCl (8 : 1 : 1), 0.3 h; (l) Pb(OAc)4, benzene, 25 *◦*C, 0.3 h, 70% (overall).

Conclusions

In conclusion, we have developed a simple and inexpensive procedure for the large-scale synthesis of both antipodes of *B*γ, γ-dimethylallyldiisopinocampheylborane, and have examined the reactivity of the reagent with a wide variety of aldehydes, providing homoallylic alcohols in high diastereo- and enantioselectivities. We have further demonstrated the application of this reagent for the synthesis of the C_1-C_6 subunit of the potent anti-cancer agent epothilone B. With an economical procedure, we believe that this reagent will find further applications in organic synthesis. We have also achieved the synthesis of the C_7-C_{21} subunit of epothilone A using a Wittig reaction and pinane-based alkoxyallyl- and crotylboration as key steps. The present study provides subunits that should be amenable to conversion into epothilone, since similar subunits have been successfully utilized previously for the synthesis of epothilone.**²***^k*

Experimental

(±)-4,4-Dimethylhex-5-en-3-ol, C8H16O, 10

A solution of dimethylallylbromide **9** (10.0 g, 67.1 mmol) was added to a suspension of zinc (8.8 g, 134.2 mmol) and propionaldehyde (9.8 mL, 134.2 mmol) in 100.0 mL THF. A saturated solution of NH₄Cl was added dropwise to the reaction mixture and stirred for 3 h at 25 *◦*C. After the completion of reaction as indicated by TLC, the reaction mixture was filtered under suction and the residue was worked up with ether and 10% HCl. The combined organic layers were dried (MgSO4), concentrated under vacuum, and purified by column chromatography (silica gel, pentane–ether, 4 : 1) to obtain 6.9 g (80%) of the homoallylic alcohol 10. $\delta_H(300 \text{ MHz}; \text{CDCl}_3;$ Me4Si) 5.80 (1 H, dd, *J* 10.8 and 17.4), 4.98–5.06 (2 H, m), 3.13 (1 H, dd, *J* 1.9 and 10.4), 1.49–1.62 (1 H, m), 1.11–1.28 (1 H, m), 0.99 (3 H, s), 0.98 (3 H, s), 0.97 (3 H, t, *J* 7.3); δ_c (75.5 MHz; CDCl3; Me4Si) 145.6, 113.1, 80.0, 41.7, 24.3, 23.1, 22.2, 11.6.

4,4-Dimethylhex-5-en-3-one, C8H14O, keto-10

Alcohol **10** (6.0 g, 46.9 mmol) was added to a suspension of Dess–Martin periodinane (29.8 g, 70.4 mmol) in 70.0 mL CH₂Cl₂ and stirred for 2 h at 25 °C. After the completion of reaction, the reaction mixture was filtered and washed with pentane. The combined organic layers were concentrated under vacuum and purified by column chromatography (silica gel, pentane–ether, 5 : 1) to obtain 5.1 g (87%) of the ketone **keto-10**. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.91 (1 H, dd, *J* 10.5 and 17.6), 5.09–5.14 (2 H, m), 2.47 (2 H, q, *J* 7.3), 1.20 (6 H, s), 0.99 (3 H, t, *J* 7.3); δ_c (75.5 MHz; CDCl₃; Me₄Si) 213.8, 142.8, 114.0, 50.7, 30.6, 23.6, 8.3.

2,2-Dimethyl-3-oxopentanal, C₇H₁₂O₂, 11

A solution of the olefinic ketone **keto-10** (5.0 g, 39.7 mmol) in CH₂CH₂–CH₂OH (1 : 1) was cooled to −78 [°]C and ozone gas was bubbled through the solution until the formation of the ozonide was complete, as indicated by the deep blue color of the solution. The resulting ozonide was quenched with $Me₂S$ (14.6 mL, 200.0 mmol) and stirred at 25 *◦*C for 4 h. The reaction mixture was concentrated under vacuum to obtain 4.0 g (80%) yield of the crude product. The resultant crude keto-aldehyde **11** was used for the next step without further purification. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 9.62 (1 H, s), 2.47 (2 H, q, *J* 7.3), 1.38 (6 H, s), 0.98 (3 H, t, *J* 7.3); δ _C(75.5 MHz; CDCl₃; Me₄Si) 209.9, 201.2, 60.2, 32.3, 19.3, 7.6.

(\pm) -4,4-Dimethyl-5-hydroxyhept-6-yn-3-one, $C_9H_{14}O_2$, OH-11

A solution of the keto-aldehyde **11** (4.0 g, 31.2 mmol) in 60.0 mL ether was cooled to 0 *◦*C and ethynylmagnesium bromide (31.5 mL, 1.0 M solution, 31.5 mmol) was added dropwise to the reaction mixture. The reaction was warmed to 25 *◦*C and stirred for 2 h. After the completion of the reaction as indicated by TLC, the reaction mixture was quenched with dilute HCl and worked up with ether. The combined organic layers were

dried (MgSO₄), concentrated under vacuum, and purified by column chromatography (silica gel, hexanes–ethyl acetate, 5 : 1) to obtain 2.9 g (69%) of the alcohol **OH-11**.

4,4-Dimethylhept-1-yn-3,5-dione, $C_9H_{12}O_2$ **, 12**

Procedure similar to that for **keto-10**, providing **12** in 87% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 3.36 (1 H, s), 2.46 (2 H, t, *J* 7.3), 1.38 (6 H, s), 1.05 (3 H, t, *J* 7.3).

$(5S)$ -4,4-Dimethyl-5-hydroxyhept-6-yn-3-one, $C_9H_{14}O_2$, 13

A solution of the diketone **12** (2.0 g, 13.2 mmol) in 25.0 mL ether was cooled to 0 *◦*C and 4.6 g (14.5 mmol) of *B*chlorodiisopinocampheylborane (DipCl) was added to it. The reaction mixture was warmed to room temperature. After the completion of the reaction as indicated by $11B$ NMR spectroscopy, the reaction mixture was oxidized with NaOAc and H_2O_2 , and worked up with ether and water. The combined organic layers were dried (MgSO₄) and concentrated under vacuum. The crude alcohol was purified by column chromatography (silica gel, hexanes–ethyl acetate, $5:1$) to obtain 1.5 g (76%) of the hydroxy-ketone **13**. The ee was determined to be 96% by HPLC analysis.

(±)-1-*tert***-Butyldimethylsilyloxy-4,4-dimethylhex-5-en-3-ol, C14H30O2Si, 16**

A solution of dimethylallylbromide **9** (10.0 g, 67.1 mmol) was added to a suspension of zinc (6.6 g, 100.0 mmol) and aldehyde **15** (6.3 g, 33.6 mmol) in 60.0 mL THF. A saturated solution of NH4Cl was added dropwise to the reaction mixture and stirred for 3 h at 25 *◦*C. After completion of the reaction as indicated by TLC, the reaction mixture was filtered under suction and the residue was worked up with ether and 10% NH₄Cl. The combined organic layers were dried (MgSO₄), concentrated under vacuum, and purified by column chromatography (silica gel, pentane–ether, $4:1$) to obtain 6.9 g (80%) of the homoallylic alcohol **16**. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.82–5.94 (1 H, m), 4.94–5.08 (2 H, m), 3.72–3.90 (2 H, m), 3.52 (1 H, m), 3.25 (1 H, br s), 1.44–1.72 (2 H, m), 1.02 (6 H, s), 0.89 (9 H, s), 0.05 (6 H, s); $δ_C(75.5 MHz; CDCl₃; Me₄Si)$ 145.7, 112.3, 78.5, 63.3, 33.4, 25.9, 22.9, 22.5, 18.2, −5.5.

1-*tert***-Butyldimethylsilyloxy-4,4-dimethylhex-5-en-3-one, C14H28O2Si, keto-16**

Procedure similar to that of **keto-10**, providing **keto-16** in 87% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.90 (1 H, dd, *J* 10.5 and 17.6), 5.12–5.18 (2 H, m), 3.85 (2 H, t, *J* 6.4), 2.67 (2 H, t, *J* 6.5), 1.22 (6 H, s), 0.87 (9 H, s), 0.04 (6 H, s); δ_c (75.5 MHz; CDCl3; Me4Si) 211.6, 142.3, 114.5, 58.9, 50.9, 40.4, 25.9, 23.2, $18.3, -5.5.$

1-Hydroxy-4,4-dimethylhex-5-en-3-one, $C_8H_{14}O_2$, 17

A solution of the ketone **keto-16** (5.5 g, 21.5 mmol) was dissolved in 40.0 mL of CH₃OH, and 10.0 mL of 10% HCl was added to the reaction mixture. After the completion of the reaction as indicated by TLC, the reaction mixture was concentrated *in vacuo* and worked up with ether and water. The combined organic layers were dried (MgSO4), and concentrated under reduced pressure. The crude hydroxyketone was purified by column chromatography (silica gel, hexanes–ethyl acetate, 4 : 1) to obtain 2.4 g (80%) of the pure hydroxyketone 17. $\delta_H(300 \text{ MHz};$ CDCl₃; Me₄Si) 5.84–5.94 (1 H, m), 5.12–5.18 (2H, m), 3.79 (2 H, t, *J* 5.4), 2.71 (2 H, t, *J* 5.0), 2.52 (1 H, br s), 1.22 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 214.3, 142.1, 114.8, 58.2, 50.9, 39.4, 23.4.

$(3S)$ -4,4-Dimethylhex-5-ene-1,3-diol, $C_8H_{16}O_2$, 18

A solution of the hydroxyketone **17** (2.4 g, 17.2 mmol) in 20.0 mL ether was cooled to 0 *◦*C, and 5.4 g (18.9 mmol) of diisopinocampheylborane ($Ipc₂BH$) was added to it. The reaction mixture was warmed to room temperature. After the completion of the reaction, as indicated by 11B NMR spectroscopy, the reaction mixture was oxidized with NaOAc and $H₂O₂$ and worked up with ether and water. The combined organic layers were dried $(MgSO₄)$ and concentrated under vacuum. The crude alcohol was purified by column chromatography (silica gel, hexanes– ethyl acetate, $5 : 1$) to obtain 1.9 g (76%) of the diol 18. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 5.84–5.94 (1 H, m), 5.12–5.18 (2 H, m), 3.75–3.85 (2 H, m), 3.54–5.64 (1 H, m), 2.9–3.1 (2 H, br s), 1.41–1.75 (2 H, m), 0.98 (6 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 145.1, 113.4, 78.3, 62.1, 41.5, 32.8, 22.8, 22.1.

(3*S***)-1-***p***-Methoxybenzyloxy-4,4-dimethylhex-5-en-3-ol, C16H24O3, PMB-18**

Diol **18** (1.8 g, 12.5 mmol) was added dropwise to a suspension of NaH (0.7 g, 50% dispersion in mineral oil, 15.0 mmol) in 30.0 mL ether at 0 *◦*C and stirred for 2 h. 6.3 g (31.2 mmol) of PMBBr was added dropwise to the reaction mixture and stirred overnight at 25 *◦*C. After the completion of the reaction as indicated by TLC, the reaction mixture was quenched with 10% HCl and worked up with ether and water. The combined organic layers were dried, concentrated under vacuum and purified by column chromatography (silica gel, hexanes–ethyl acetate, 5 : 1), to obtain 2.6 g (80%) of mono-protected alcohol **PMB-18**. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 7.25 (1 H, d, *J* 8.2), 6.88 (2 H, d, *J* 8.6), 5.86 (1 H, dd, *J* 6.1 and 11.1), 4.99–5.06 (2 H, m), 4.45 (2 H, s), 3.79 (3 H, s), 3.48–3.73 (3 H, m), 2.89 (1 H, br s), 1.55–1.92 (2 H, m), 1.05 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 159.3, 145.6, 130.1, 129.3, 113.9, 112.6, 77.9, 73.0, 69.7, 55.3, 41.3, 31.3, 22.9, 22.5.

(3*S***)-3-***tert***-Butyldimethylsilyloxy-1-***p***-methoxybenzyloxy-4,4** dimethylhex-5-ene, C₂₂H₃₈O₃Si, 19

Alcohol **PMB-18** (2.5 g, 9.5 mmol) was dissolved in 38 mL dimethylformamide. *tert*-Butyldimethylsilylchloride (1.7 g, 11.3 mmol) and imidazole (1.2 g, 17.0 mmol) were added at 0 *◦*C and the mixture stirred for 3 h. After the completion of the reaction as indicated by TLC, the reaction mixture was worked up with ether and water. The organic layer was dried over MgSO4, concentrated under aspirator vacuum to obtain 2.8 g (80%) of silyl ether 19. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 7.25 (2 H, d, *J* 7.9), 6.88 (2 H, d, *J* 8.7), 5.86 (1 H, dd, *J* 10.4 and 18.0), 4.93–4.98 (2 H, m), 4.40 (2 H s), 3.81 (3 H, s), 3.42–3.52 (3 H, m), 1.53–1.93 (2 H, m), 0.97 (6 H, s), 0.89 (9 H, s), 0.04 (3 H, s), 0.03 (3 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 159.1, 146.1, 130.8, 129.3, 113.8, 111.6, 76.4, 72.5, 67.8, 55.3, 42.5, 33.9, 26.2, 24.5, 22.6, 18.4, −3.8; EI-MS: 309 $[(M - C(CH_3)_2CH=CH_2)^+]$, 121 $[100\%, {}^+CH_2C_6H_4OCH_3]$, 73, 59; CI-MS: 379 $[(M + H)^{+}]$, 121 [100%, CH₂C₆H₄OCH₃⁺], 81, 69; HRMS-CI: (M⁺) 378.2608 (actual), 378.2590 (calcd); $(M + H)^+$ 379.2662 (actual), 379.2669 (calcd).

(3*S***)-3-***tert***-Butyldimethylsilyloxy-5-***p***-methoxybenzyloxy-2,2** dimethylpentanal, C₂₁H₃₆O₄Si, ald-19

Procedure similar to that of **11**. 78% pure aldehyde **ald-19** was obtained. The resulting aldehyde was used for the next step without further purification. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 9.54 (1 H, s), 7.25 (2 H, d, *J* 8.5), 6.88 (2 H, d, *J* 8.5), 4.40 (2 H, s), 3.96 (1 H, dd, *J* 3.51 and 7.41), 3.81 (3 H, s), 3.48 (2 H, dd, *J* 6.0 and 7.4), 1.61–1.91 (2 H, m), 1.05 (3 H, s), 1.00 (3 H, s), 0.85 (9 H, s), 0.05 (3 H, s), 0.03 (3 H, s); δ_c (75.5 MHz; CDCl₃; Me4Si) 206.3, 159.2, 130.4, 129.3, 113.8, 73.3, 72.6, 66.7, 55.3, 51.3, 33.6, 26.0, 19.1, 18.2, 17.6, −4.0, −4.2.

(5*S***)-5-***tert***-Butyldimethylsilyloxy-7-***p***-methoxybenzyloxy-4,4 dimethylheptan-3-ol, C23H42O4Si, OH-19**

Aldehyde **ald-19** (1.0 g, 2.6 mmol) was dissolved in 5.0 mL THF and cooled to 0 *◦*C. Ethylmagnesium bromide (2.9 mL, 1.0 M solution in THF, 2.9 mmol) was added to it dropwise and stirred for 1 h. The reaction mixture was quenched with sat. NH₄Cl, worked up with ether and the combined organic layers were dried (MgSO₄), and concentrated under vacuum to obtain crude alcohol **OH-19**, which was used for the next step without further purification.

(5*S***)-5-***tert***-Butyldimethylsilyloxy-7-***p***-methoxybenzyloxy-4,4** dimethylheptan-3-one, C₂₃H₄₀O₄Si, 2

Procedure similar to that for **keto-10**, providing 2 (the C_1-C_6 subunit of epothilone) in 88% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.24 (2 H, d, *J* 8.6), 6.87 (2 H, d, *J* 8.6), 4.40 (2 H, s), 4.05 (1 H, dd, *J* 3.1 and 7.4), 3.80 (3 H, s), 3.42–3.47 (2 H, m), 2.39–2.61 (2 H, m), 1.50–1.76 (2 H, m), 1.10 (3 H, s), 1.05 (3 H, s), 0.99 (3 H, t, *J* 7.1), 0.86 (9 H, s), 0.04 (3 H, s), 0.01 (3 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 159.2, 130.6, 129.2, 113.8, 74.0, 72.5, 67.3, 55.3, 53.0, 34.3, 31.7, 26.1, 22.0, 20.2, 18.3, 7.8, −4.0, −4.1; EI-MS: *m*/*z* 309 [(M − C(CH3)2COCH2CH3) +], 121 $[100\%, ^+CH_2C_6H_4OCH_3]$, 77, 75; CI-MS: 409 $[(M + H)⁺]$, 271, 121 [100%, +CH2C6H4OCH3], 81, 69; HRMS-CI: (M+) 408.2696 (calcd), 408.2677 (actual); $(M + H)^+$ 409.2774 (calcd), 409.2787 (actual).

*B***-c,c-Dimethylallyldiisopinocampheylborane 20**

1-Chloro-3-methyl-2-butene (90.2 mL, 1.0 mol) was added dropwise to a stirred suspension of magnesium (120 g, 2.5 mol) in 200 mL ether and was stirred for 1 h. Meanwhile, a flame-dried round-bottomed flask was cooled under an inert atmosphere and *B*-methoxydiisopinocampheylborane (158.2 g, 0.5 mol) was weighed into the flask under nitrogen. The borinate was dissolved in 500 mL ether and cooled to 0 *◦*C. The previously made Grignard reagent was then transferred into the borinate solution and stirred for 4 h. After the completion of the reaction as monitored by $11B$ NMR (δ 79), the reaction mixture was filtered under nitrogen using a Kramer's filter and was washed repeatedly with ether. The organic layer was concentrated under vacuum in nitrogen atmosphere. Spectroscopic grade pentane was added to it using a cannula, stirred for 5 min and the precipitate (unreacted Grignard reagent and the magnesium salts) allowed to settle. The supernatant liquid was then transferred *via* a cannula into another round-bottomed flask under nitrogen and the solvent was evaporated off under vacuum. After repeated washing with pentane, the concentrate (90%, 159.4 g, 450.0 mmol) was dissolved in 450 mL pentane so as to prepare a 1 M stock solution of the reagent in pentane. This stock solution was used for reaction with various aldehydes.

Typical procedure for the reaction of **20** with an aldehyde: *B*- γ , γ-dimethylallyldiisopinocampheylborane, **20** (24.0 mL, 1.0 M solution in pentane, 24.0 mmol) was dissolved in 24.0 mL ether and cooled to −100 *◦*C. The aldehyde (20.0 mmol) was dissolved in 10.0 mL ether, cooled to −78 *◦*C, and transferred to the reaction mixture using a cannula. The reaction mixture was monitored using 11B NMR (*d* 56). After completion, the reaction mixture was oxidized using 3.0 M NaOH and 30% H_2O_2 and stirred at room temperature for 6 h. The reaction mixture was worked up with ether and water, the combined organic layers were dried (MgSO₄), concentrated under vacuum, and purified by column chromatography (silica gel, hexanes–ethyl acetate) to afford pure homoallylic alcohol.

(3*S***)-1-***p***-Methoxybenzyloxy-4,4-dimethylhex-5-en-3-ol,** $C_{16}H_{24}O_3$, 24f

Aldehyde **23f** (7.8 g, 40.4 mmol) was added to a stirred solution of (−)-*B*-γ,γ-dimethylallyldiisopinocampheylborane (96.9 mL

of a 0.5M solution in ether–pentane) at −100 *◦*C and maintained at that temperature for 2 h. The progress of the reaction was followed by ¹¹B NMR spectroscopy (δ 56). Upon completion, the mixture was oxidized with 19.4 mL of 3.0 M NaOH and 19.4 mL of 30% H_2O_2 , stirred for 4 h at room temperature and extracted with ether. The organic layer was dried over MgSO₄, concentrated under vaccum, purified by column chromatography (silica gel, hexanes–ethyl acetate, 3 : 2) to obtain 8.7 g (82%) of the pure alcohol 24f. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.25 (2 H, d, *J* 8.2), 6.88 (2 H, d, *J* 8.6), 5.86 (1 H, dd, *J* 6.1 and 11.1), 4.99–5.06 (2 H, m), 4.45 (2 H, s), 3.79 (3 H, s), 3.48–3.73 (3 H, m), 2.89 (1 H, d, *J* 2.7), 1.55–1.92 (2 H, m), 1.05 (6 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 159.3, 145.6, 130.1, 129.4, 129.3, 113.9, 112.6, 77.9, 73.0, 69.7, 55.3, 41.3, 31.3, 22.9, 22.5.

(2*S***,3***S***)-1-***p***-Methoxybenzyloxy-2,4,4-trimethylhex-5-en-3-ol, C17H26O3, 24g**

Procedure as described above for the dimethylallylboration of aldehydes. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.23 (2 H, d, *J* 8.3), 6.87 (2 H, d, *J* 8.5), 5.98 (1 H, dd, *J* 6.8 and 10.8), 4.95–5.08 (2 H, m), 4.41 (2 H, dd, *J* 6.3 and 7.1), 3.79 (3 H, s), 3.20–3.60 (4 H, m), 1.91–2.07 (1 H, m), 1.05 (3 H, s), 1.05 (3 H, s), 0.94 (3 H, d, *J* 6.9); δ _C(75.5 MHz; CDCl₃; Me₄Si) 159.3, 145.9, 129.3, 129.2, 113.8, 111.6, 83.2, 79.2, 76.5, 73.7, 73.0, 55.3, 42.3, 34.6, 24.2, 23.8, 18.4, 11.5.

(2*S***)-1-Benzyloxypent-4-en-2-ol, C12H16O2, Bn-OH-23h**

To a solution of *B*-allyldiisopinocampheylborane (20.0 mL, 1.0 M solution, 20.0 mmol) in ether–pentane was added benzyloxyacetaldehyde (3.3 g, 22.0 mmol), and the mixture stirred for 3 h at −100 *◦*C. After the completion of the reaction as indicated by ¹¹B NMR spectroscopy $(\delta 79)$, the reaction mixture was oxidized using 8.0 mL of 3.0 M NaOH and 8.0 mL of 30% H_2O_2 , and the reaction mixture worked up with ether and water. The combined organic layers were dried, concentrated and used in the next step without further purification. $\delta_H(300 \text{ MHz};$ CDCl3; Me4Si) 7.30–7.37 (5 H, m), 5.77–5.90 (1 H, m), 5.08– 5.17 (2 H, m), 4.57 (2 H, s), 3.85–3.93 (1 H, m), 3.52 (1 H, dd, *J* 3.4 and 9.4), 3.38 (1 H, dd, *J* 7.4 and 9.5), 2.24–2.30 (3 H, m); δ _C(75.5 MHz; CDCl₃; Me₄Si) 138.0, 134.3, 128.5, 127.8, 127.7, 117.8, 73.9, 73.4, 69.8, 37.9.

(2*S***)-2-***tert***-Butyldimethyldimethylsilyloxy-1-benzyloxypent-4 ene, C18H30O2, Bn-TBS-23h**

Procedure similar to that for **19**, providing the pure silyl ether **Bn-TBS-23h** in 82% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 7.30– 7.40 (5 H, m), 5.73–5.91 (1 H, m), 5.05–5.20 (2 H, m), 4.56 (2 H, s), 3.86–3.97 (1 H, m), 3.38 (2 H, s), 2.24–2.30 (2 H, m), 0.90 $(9 \text{ H}, \text{s})$, 0.1 (6 H, s); $\delta_c(75.5 \text{ MHz}; \text{CDCl}_3$; Me₄Si) 138.5, 135.0, 128.4, 127.6, 127.5, 117.1, 74.3, 73.4, 71.3, 39.5, 25.9, 18.3, −4.4, −4.6.

(2*S***)-2-***tert***-Butyldimethyldimethylsilyloxypent-4-en-1-ol, C11H24O2, OH-TBS-23h**

The silylated benzyl ether **Bn-TBS-23h** (1.6 g, 5.9 mmol) was dissolved in 12.0 mL THF and transferred to a pre-cooled solution of lithium (0.2 g, 28.6 mmol) in liquid ammonia (60.0 mL) and stirred for 3 h at −78 *◦*C. The reaction mixture was slowly quenched with NH4Cl and MeOH, warmed to room temperature, stirred overnight and worked up with ether and water. The organic layer was dried $(MgSO₄)$, concentrated under aspirator vacuum, purified by column chromatography (silica gel, hexanes–ethyl acetate, 5 : 1) to obtain 0.8 g (68%) of alcohol **OH-TBS-23h**. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 5.65–5.86 (1 H, m), 4.94–5.15 (2 H, m), 3.37–3.78 (4 H, m), 2.24–2.30 (2 H, m), 0.90 (9 H, s), 0.10 (6 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 134.5, 117.4, 71.2, 66.5, 37.6, 25.9, 18.3, −5.3, −5.4.

(2*S***)-Pent-4-ene-1,2-diol, C5H10O2, diol-23h**

A solution of the silyl alcohol **OH-TBS-23h** (0.5 g, 2.7 mmol) was dissolved in 6.0 mL CH₃OH, and 1.5 mL of 10% HCl added. The reaction was stirred for 2 h, and monitored by TLC. The reaction mixture was concentrated under vacuum, and worked up with ether and water. The combined organic layers were dried, concentrated and purified by column chromatography (silica gel, hexanes–ethyl acetate, 2 : 3) to obtain 0.23 g (84%) of the diol **diol-23h**. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.71–5.85 (1 H, m), 5.05– 5.11 (2 H, m), 3.83 (2 H, br s), 3.70 (1 H, ddd, *J* 2.9, 6.6 and 13.8), 3.59 (1 H, dd, *J* 2.9 and 11.5), 3.40 (1 H, dd, *J* 7.6 and 11.2), 2.19 (2 H, t, *J* 6.9); δ_c (75.5 MHz; CDCl₃; Me₄Si) 134.2, 117.8, 71.5, 66.1, 37.8; EI-MS: *m*/*z* 85 [M + H − H2O], 61, 43; CI-MS: 103 $[(M + H)^+]$, 85 $[(M + H - H_2O)^+]$, 67 $[100\%, M +$ $H - 2H₂O$].

$(2S)$ -1,2-Dimethoxypent-4-ene, $C_7H_{14}O_2$, diMe-23h

A solution of **diol-23h** (0.5 g, 5.0 mmol) in 10.0 mL ether was added dropwise to a pre-cooled suspension of NaH (0.36 g, 50% dispersion in mineral oil, 7.5 mmol) in ether. CH₃I (2.1 g, 15.0 mmol) was added to the reaction mixture and stirred overnight at 25 *◦*C. After the completion of the reaction, the reaction mixture was quenched with dilute HCl and worked up ether and water. The combined organic layers were dried over MgSO4 and concentrated under vacuum. The crude product was purified by column chromatography (silica gel, hexanes– ethyl acetate, 4 : 1) to obtain 0.48 g (74%) of the dimethyl ether **diMe-23h**.

$(3S)$ -3,4-Dimethoxybutanal, $C_6H_2O_3$, 23h

Procedure similar to that for **11**, to provide the aldehyde **23h** in 68% yield.

$(2S.4S)$ -1, 2-Dimethoxy-5,5-dimethylhept-6-en-4-ol, $C_1H_2O_3$, **24h**

Procedure as described earlier for the dimethylallylboration of aldehydes. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 5.85 (1 H, dd, *J* 5.7 and 11.1), 4.97–5.04 (2 H, m), 3.35–3.55 (5 H, m), 3.43 (3 H, s), 3.36 (3 H, s), 1.67–1.74 (1 H, m), 1.44–1.54 (1 H, m), 1.01 (6 H, s); $δ_c(75.5 MHz; CDCl₃; Me₄Si)$ 145.5, 112.4, 80.9, 77.5, 74.2, 59.2, 57.4, 41.3, 33.3, 23.1, 22.1; EI-MS: *m*/*z* 157, 133, 125, 101 [100%], 89, 69, 59; CI-MS: 203 [M + H], 185 [(M + $H - H₂O$ ⁺], 153, 133, 121, 101; HRMS-CI: $(M + H - H₂O)$ ⁺ 185.1547 (actual), 185.1542 (calcd).

1-*tert*-Butyldimethylsilyloxy-2-methylhex-5-ene, C₁₃H₂₈OSi, 31

A solution of the iodide **30** (10.0 g, 3.2 mmol) was dissolved in 6.0 mL CH₂Cl₂ and cooled to −78 °C. Lithium tetrachlorocuprate (16.0 mL, 0.1 M solution, 1.6 mmol) and allylmagnesium bromide (6.4 mL, 1 M solution, 6.4 mmol) were added to the reaction mixture and stirred at −78 *◦*C for 4 h. After complete reaction of the starting material, the reaction mixture was worked up with NH4Cl and ether. The combined organic layers were dried, concentrated *in vacuo*, and purified by column chromatography (silica gel, hexanes–ethyl acetate, 9 : 1) to obtain 3.6 g (50%) of the olefin 31. $\delta_H(300 \text{ MHz}; \text{CDCl}_3;$ Me4Si) 5.77–5.86 (1 H, m), 5.94–5.03 (2 H, m), 3.35–3.48 (2 H, m), 1.96–2.17 (2 H, m), 1.44–1.65 (2 H, m), 1.09–1.22 (1 H, m), 0.90 (9 H, s), 0.88 (3 H, d, *J* 6.5), 0.05 (6 H, s); δ_c (75.5 MHz; CDCl3; Me4Si) 139.3, 114.2, 68.2, 35.3, 32.2, 25.9, 19.0, 18.4, 16.6, −5.3; EI-MS: *m*/*z* 213 [M − CH3], 171 [M − C4H9], 141, 115, 99, 89, 75 [100%, (HOSi(CH3)2) +]; CI-MS: *m*/*z* 229 [(M + H)⁺], 213 [M + H – CH₄], 171, 145, 133, 115; HRMS-CI: (M + H)+ 229.1982 (actual), 229.1988 (calcd).

6-*tert***-Butyldimethylsilyloxy-5-methylhexan-1-ol, C13H30O2Si, OH-31**

Olefin **31** (3.5 g, 15.4 mmol) was dissolved in 30.0 mL ether. BH_3 ·Me₂S (1.2 mL, 10 M solution, 12.0 mmol) was added at 0 *◦*C and the mixture stirred overnight at room temperature. The reaction mixture was oxidized with 14.4 mL of 3.0 M sodium hydroxide and 14.4 mL of 30% hydrogen peroxide and stirred for 10 h at 25 *◦*C. The reaction mixture was worked up with ether and water. The organic layer was dried over $MgSO₄$, and evaporated under aspirator vacuum to obtain 1.5 g (40%) of alcohol **OH-31**. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 3.63 (2 H, t, *J* 7.0), 3.42 (1 H, dd, *J* 6.0 and 11.0), 3.35 (1 H, dd, *J* 7.0 and 11.0), 1.57–1.53 (3 H, m), 1.42–1.39 (3 H, m), 1.16–1.06 (1 H, m), 0.88 (9 H, s), 0.85 (3 H, d, *J* 6.5), 0.03 (3 H, s), 0.02 (3 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 68.2, 62.7, 35.6, 32.9, 32.8, 25.7, 23.0, 18.2, 16.5, −5.5; EI-MS: *m*/*z* 189, 171, 97, 75, 55 [100%, C₄H₇⁺], 47; CI-MS: *m*/*z* 247 [(M + H)⁺], 231, 189, 155, 143, 133, 115 [M + H – HOSi(CH₃)₂C(CH₃)₃]; HRMS-CI: (M + H)+ 247.2088 (actual), 247.2093 (calcd).

6-*tert***-Butyldimethylsilyloxy-1-iodo-5-methylhexane, C13H29OISi, 25**

Alcohol **OH-31** (1.5 g, 6.1 mmol) was dissolved in 12.0 mL $CH₂Cl₂$. Iodine (3.1 g, 12.2 mmol), triphenylphosphine (3.2 g, 12.2 mmol) and imidazole (0.8 g, 12.2 mmol) were added to the reaction mixture, which was stirred for 8 h. After the completion of the reaction as indicated by TLC, the reaction mixture was worked up with ether and water. The organic layers were dried, concentrated under vacuum and purified by column chromatography (silica gel, hexanes–ethyl acetate, 19 : 1) to afford 1.7 g (80%) of 25. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 3.42 (1) H, dd, *J* 6.5 and 10.0), 3.38 (1 H, dd, *J* 6.0 and 10.0), 3.19 (2 H, t, *J* 7.0), 1.85–1.78 (2 H, m), 1.61–1.55 (1 H, m), 1.47– 1.33 (3 H, m), 1.10–1.02 (1 H, m), 0.89 (9 H, s), 0.87 (3 H, d, *J* 6.7), 0.04 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 68.1, 35.4, 33.7, 31.8, 27.8, 25.8, 18.2, 16.5, 7.1, −5.5; EI-MS: *m*/*z* 299 [M − C4H9], 215, 185, 97 (100%), 75, 55; CI-MS: *m*/*z* 357 [(M + H)⁺], 341, 299, 229 [M + H – HI], 133 [(HOSi(CH₃)₂C(CH₃)₃ + H)⁺], 115; HRMS-CI: $(M + H)$ ⁺ 357.1103 (actual), 357.1111 (calcd).

$(2S,3S)$ -3- $(2-Methoxyethoxymethoxy)$ pent-4-en-2-ol, $C_9H_{18}O_4$, **33**

sec-Butyllithium (193.0 mL, 231.7 mmol) was added dropwise to a well-stirred solution of allyloxymethoxyethoxymethane **32** (35.6 g, 243.8 mmol) in 500.0 mL THF at −78 *◦*C and stirred for 0.5 h. $(+)$ -Ipc₂BOMe $((+)$ -*B*-methoxydiisopinocampheylborane) (92.6 g, 292.7 mmol, 1.0 M solution in THF) was added to the reaction mixture and stirred for 1 h at −78 °C. BF₃·Et₂O (55.6 mL, 439.0 mmol) was added and the reaction mixture was cooled to−100 *◦*C. Acetaldehyde (13.7 mL, 243.9 mmol) dissolved in 25 mL THF was cooled to −78 *◦*C and was added dropwise to the reaction mixture. The reaction was stirred at −100 *◦*C for 2 h and then warmed to −78 *◦*C and stirred overnight. The reaction mixture was oxidized with 117.2 mL of 3.0 M sodium hydroxide and 117.2 mL of 30% hydrogen peroxide and stirred for 10 h at room temperature. The product was extracted with ether, washed with water, and dried over MgSO4. The solvent was removed under aspirator vacuum. The byproducts isopinocampheol and α -pinene were vacuum-distilled and the crude product was purified by silica gel column chromatography (hexane–ethyl acetate, 7 : 3) to obtain 33.0 g (75%) of the homoallylic alcohol 33. $\delta_H(300 \text{ MHz}; \text{CDCl}_3;$ Me4Si) 5.64 (1 H, ddd, *J* 7.8, 10.6 and 17.8), 5.27–5.32 (2 H, m), 4.78 (1 H, d, *J* 7.0), 4.69 (1 H, d, *J* 7.0), 3.78–3.86 (2 H, m), 3.63–3.73 (2 H, m), 3.53–3.56 (2 H, m), 3.38 (3 H, s), 2.85 (1 H, br s), 1.13 (3 H, d, *J* 6.3); $\delta_c(75.5 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 134.7,

119.6, 92.9, 82.9, 71.7, 69.3, 67.3, 58.8, 18.5; EI-MS: *m*/*z* 89 [100%, +CH2OCH2CH2OCH3], 77, 70, 59, 45, 43; CI-MS: *m*/*z* 191 $[(M + H)^+]$, 115.

$(3S)$ -3- $(2-Methoxyethoxymethoxy)$ pent-4-en-2-one, $C_9H_{16}O_4$, **35**

Procedure similar to that for **keto-10**, providing the ketone **35** in 76% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.77 (1 H, ddd, *J* 6.6, 10.3 and 17.1), 5.36–5.51 (2 H, m), 4.80 (1 H, d, *J* 6.9), 4.76 (1 H, d, *J* 6.9), 4.60 (1 H, d, *J* 6.6), 3.65–3.80 (2 H, m), 3.53 (2 H, t, *J* 4.4), 3.37 (3 H, s), 2.19 (3 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 206.1, 132.3, 119.8, 93.6, 82.8, 71.6, 67.4, 58.8, 25.7; EI-MS: m/z 145 [(M − CH₃CO)⁺], 113, 89 [100%, CH2OCH2CH2OCH3 +], 77, 59, 43; CI-MS: *m*/*z* 189 $[(M + H)^+]$, 113, 89 [100%, $+CH_2OCH_2CH_2OCH_3$].

(3*S***,4***S***)-3-(2-Methoxyethoxymethoxy)pentane-1,4-diol,** $C_9H_{20}O_5$, 37

Olefin **33** (5.0 g, 26.3 mmol) was added to a suspension of dicyclohexylborane (11.7 g, 65.8 mmol) in 60.0 mL ether at 0 *◦*C and stirred overnight at room temperature. The reaction mixture was oxidized with 26.2 mL of 3.0 M sodium hydroxide and 26.2 mL of 30% hydrogen peroxide and stirred for 10 h at room temperature. The reaction mixture was worked up with ether and water. The organic layer was dried over MgSO4, and evaporated under aspirator vacuum to obtain 4.1 g (76%) of alcohol **37**. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 4.72–4.84 (2 H, m), 3.62–3.78 (5 H, m), 3.46–3.58 (3 H, m), 3.37 (3 H, s), 2.90–3.10 $(2 H, br s), 1.72-1.86 (1 H, m), 1.54-1.68 (1 H, m), 1.14 (3 H, d,$ J 6.2); δ_c (75.5 MHz; CDCl₃; Me₄Si) 96.5, 82.1, 71.7, 69.6, 67.7, 59.0, 58.7, 33.9, 19.0.

(2*S***,3***S***)-5-***tert***-Butyldimethylsilyloxy-3-(2 methoxyethoxymethoxy)pentan-2-ol, C15H34O5Si, TBS-37**

Procedure similar to that for **Bn-TBS-23h**, providing the silyl ether **TBS-37** in 87% yield. $δ_H(300 MHz, CDCl_3; Me_4Si)$ 4.80 (1 H, d, *J* 6.9), 4.76 (1 H, d, *J* 6.9), 3.45–3.82 (8 H, m), 3.36 (3 H, s), 1.73–1.83 (1 H, m), 1.53–1.64 (1 H, m), 1.14 (3 H, d, *J* 6.4), 0.86 $(9 \text{ H}, \text{s})$, 0.02 (6 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 96.5, 82.8, 71.7, 69.4, 67.7, 59.1 (2 carbons), 34.4, 25.9, 18.9, 18.2, −5.3, $-5.4.$

(3*S***)-5-***tert***-Butyldimethylsilyloxy-3-(2** methoxyethoxymethoxy)pentan-2-one, C₁₅H₃₂O₅Si, 38

Procedure similar to that for **keto-10**, providing the ketone **38** in 80% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 4.72 (2 H, s), 4.18 (1 H, dd, *J* 4.5 and 7.5), 3.66–3.73 (4 H, m), 3.46–3.49 (2 H, m), 3.35 (3 H, s), 2.15 (3 H, s), 1.78–1.87 (2 H, m), 0.86 (9 H, s), 0.02 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 209.1, 95.6, 79.5, 71.7, 67.6, 59.0, 58.5, 35.0, 26.4, 25.9, 18.9, −5.4, −5.5.

4-((3*S***)-5-***tert***-Butyldimethylsilyloxy-3-methoxyethoxymethoxy-**2-methylpent-1-enyl)-2-methylthiazole, C₂₀H₃₇O₄SNSi, 39

Thiazolyl chloride **5** (2.1 g, 14.1 mmol) was heated at 80 *◦*C with tri-*n*-butylphosphine (42.3 mL of a 1.0 M solution, 42.3 mmol) for 3 h. The reaction mixture was cooled to 0 *◦*C and diluted with 30.0 mL THF. NaHMDS (13.4 mL, 1.0 M solution, 13.4 mmol) was added dropwise to the reaction mixture and stirred at 0 *◦*C for 20 min. Ketone **38** (3.0 g, 9.4 mmol) was added to the reaction mixture and stirred overnight. After the completion of the reaction as indicated by TLC, the reaction mixture was quenched with NH4Cl and worked up with ether and water. The combined organic layers were dried, concentrated and purified by column chromatography (silica gel, hexanes–ethyl acetate, 9 : 1) to obtain 2.9 g (74%) of the coupled olefin 39. $\delta_H(300 \text{ MHz};$ CDCl3; Me4Si) 6.89 (1 H, s), 6.45 (1 H, s), 4.68 (1 H, d, *J* 6.7), 4.59 (1 H, d, *J* 6.8), 4.25 (1 H, dd, *J* 5.4 and 8.0), 3.47–3.80 (6 H,

m), 3.32 (3 H, s), 2.65 (3 H, s), 1.95 (3 H, s), 1.67–1.91 (2 H, m), 0.84 (9 H, s), 0.01 (6 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 164.4, 152.9, 138.5, 121.3, 115.9, 92.8, 78.5, 71.8, 67.0, 59.7, 59.0, 37.3, 25.9, 19.2, 18.2, 13.6, −5.3, −5.4.

4-((3*S***)-5-Hydroxy-3-(2-methoxyethoxymethoxy)-2-methylpent-**1-enyl)-2-methylthiazole, C₁₄H₂₃O₄SN, 40

A solution of the silyl ether **39** (2.8 g, 6.7 mmol) was dissolved in 10.0 mL of CH₃OH, and 2.0 mL of AcOH was added to the reaction mixture. After the completion of the reaction as indicated by TLC, the reaction mixture was concentrated *in vacuo* and worked up with ether and water. The combined organic layers were dried $(MgSO₄)$, and concentrated under reduced pressure. The crude hydroxyketone was purified by column chromatography (silica gel, hexanes–ethyl acetate, 4 : 1) to obtain 1.7 g (83%) of the pure alcohol 39. $\delta_H(300 \text{ MHz};$ CDCl3; Me4Si) 6.85 (1 H, s), 6.43 (1 H, s), 4.63 (1 H, d, *J* 6.9), 4.53 (1 H, d, *J* 6.9), 4.30 (1 H, dd, *J* 4.7 and 8.9), 3.45–3.83 (6 H, m), 3.30 (3 H, s) 2.60 (3 H, s), 1.92 (3 H, s), 1.71–1.79 (2 H, m); δ _C(75.5 MHz; CDCl₃; Me₄Si) 164.6, 152.6, 138.5, 120.8, 116.0, 92.3, 78.7, 71.8, 67.0, 59.1, 59.0, 36.8, 19.2, 14.0.

4-((3*S***)-5-Oxo-3-(2-methoxyethoxymethoxy)-2-methylpent-1** enyl)-2-methylthiazole, C₁₄H₂₁O₄SN, 27

Procedure similar to that for **keto**-**10**, providing the aldehyde **27** in 85% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 9.77 (1 H, s), 6.93 (1 H, s), 6.53 (1 H, s), 4.67–4.72 (2 H, m), 4.60 (1 H, d, *J* 6.9), 3.47–3.79 (4 H, m), 3.34 (3 H, s), 2.77 (1 H, ddd, *J* 2.9, 9.3 and 16.2), 2.66 (3 H, s), 2.52 (1 H, ddd, *J* 1.7, 4.0 and 16.2), 2.01 (3 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 200.7, 164.8, 152.3, 136.6, 121.8, 116.7, 92.6, 76.2, 71.7, 67.3, 59.0, 47.8, 19.2, 13.9.

4-((1*E***,3***Z***,5***Z***,10***S***)-11-***tert***-butyldimethylsilyloxy-2,10** dimethylundeca-1,3,5-trienyl)-2-methylthiazole, C₂₃H₃₉OSNSi, 42

A mixture of iodide $25(4.7 \text{ g}, 13.1 \text{ mmol})$ and $Ph_3P(3.8 \text{ g},$ 14.4 mmol, 1.1 equiv.) was heated neat at 100 *◦*C for 2 h to provide the phosphonium salt (7.4 g, 91%) as a white solid: The salt was dissolved in THF (120 mL, 0.1 M), and the solution was cooled to 0 *◦*C. Sodium hexamethyldisilylamide (NaHMDS, 12.0 mL, 12.0 mmol, 1 M solution in THF) was slowly added at the same temperature, and the resulting mixture was stirred for 15 min, before aldehyde **27** (3.2 g, 9.8 mmol) was slowly added. Stirring was continued for another 15 min at 0 *◦*C, and then the mixture was quenched with saturated aqueous $NH₄Cl$. The reaction mixture was worked up with ether and water. The combined organic layers were dried (MgSO4), and concentrated under reduced pressure to afford the crude product. Purification by column chromatography (silica gel, hexane–ethyl acetate, 9 : 1) furnished olefin 42 with extensive β -elimination.

(2*S***,3***S***)-2-***tert***-Butyldimethylsilyloxy-3-(2** methoxyethoxymethoxy)pent-4-ene, C₁₅H₃₂O₄Si, 45

Procedure similar to that for **19**, providing the silyl ether **45** in 89% vield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.73 (1 H, ddd, *J* 6.8, 10.4 and 17.3), 5.20–5.27 (2 H, m), 4.74 (1 H, d, *J* 6.9), 4.70 (1 H, d, *J* 6.9), 3.73–3.94 (3 H, m), 3.58–3.64 (1 H, m), 3.50–3.54 (2 H, m), 3.36 (3 H, s), 1.07 (3 H, d, *J*), 0.86 (9 H, s), 0.04 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 134.9, 118.2, 93.6, 81.2, 71.8, 70.3, 66.8, 59.0, 25.9, 19.2, 18.1, −4.6, −4.7.

(3*S***,4***S***)-4-***tert***-Butyldimethylsilyloxy-3-(2** methoxyethoxymethoxy)pentan-1-ol, C₁₅H₃₄O₅Si, OH-45

Procedure similar to that for **37**, providing the alcohol **OH-45** in 78% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 4.83 (1 H, d, *J* 6.9), 4.73 (1 H, d, *J* 6.9), 3.53–3.90 (8 H, m), 3.38 (3 H, s), 3.00 (1 H, br s), 1.86 (1 H, ddd, *J* 4.9, 9.6 and 14.4), 1.50–1.61 (1 H, m), 1.11 (3 H, d, *J* 6.3), 0.87 (9 H, s), 0.07 (3 H, s), 0.06 (3 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 96.1, 80.0, 71.7, 70.2, 67.4, 59.4, 59.1, 32.6, 25.8, 18.3, 18.0, −4.7, −4.8.

(3*S***,4***S***)-4-***tert***-Butyldimethylsilyloxy-3-(2** methoxyethoxymethoxy)pentanal, C₁₅H₃₂O₅Si, 46

Procedure similar to that for **keto-10**, providing the pure aldehyde **46** in 97% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 9.72 (1 H, s), 4.71 (2 H,s), 3.95–4.04 (2 H, s), 3.59–3.62 (2 H, m), 3.47–3.49 (2 H, m), 3.33 (3 H, s), 2.43–2.68 (2 H, m), 1.05 (3 H, d, *J* 6.2), 0.81 (9 H, s), 0.01 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me4Si) 201.2, 95.7, 76.5, 71.6, 68.6, 67.2, 59.0, 43.8, 25.7, 17.9, $17.5, -4.7, -4.9.$

(2*S***,3***S***,5***Z***,10***S***)-2,11-Bis(***tert***-butyldimethylsilyloxy)-3-(2** methoxyethoxymethoxy)undec-5-ene, C₂₈H₆₀O₅Si₂, 47

Procedure similar to that for **42**, providing the olefin **47** in $\lt 5\%$ yield.

(2*S***,3***S***)-2-Benzyloxy-3-(2-methoxyethoxymethoxy)pent-4-ene, C16H24O4, Bn-33**

A solution of homoallylic alcohol **33** (10.0 g, 52.6 mmol) was added dropwise to a suspension of NaH (3.8 g, 50% dispersion in mineral oil, 79.0 mmol) in 100.0 mL ether at 0 *◦*C. Benzyl bromide (12.0 mL, 105.2 mmol) was added to the reaction mixture and stirred overnight at room temperature. The reaction mixture was quenched with NH4Cl and worked up with ether and water. The combined organic layers were dried $(MgSO₄)$, concentrated under reduced pressure and purified by column chromatography (silica gel, hexanes–ethyl acetate 19 : 1) to obtain 11.5 g (78%) of the benzyl ether **Bn-33**. $\delta_H(300 \text{ MHz};$ CDCl3; Me4Si) 7.24–7.42 (5 H, m), 5.81 (1 H, ddd, *J* 7.2, 10.3 and 17.6), 5.24–5.34 (2 H, m), 4.79 (1 H, d, *J* 6.8), 4.73 (1 H, d, *J* 6.9), 4.65 (1 H, d, *J* 12.0), 4.60 (1 H, d, *J* 11.9), 4.14 (1 H, dd, *J* 5.7 and 6.4), 3.78–3.85 (1 H, m), 3.45–3.65 (4 H, m), 3.34 (3 H, s), 1.18 (3 H, d, *J* 6.3); δ_c (75.5 MHz; CDCl₃; Me₄Si) 138.9, 134.9, 128.3, 127.8, 127.5, 118.6, 93.2, 79.8, 76.8, 71.8, 71.6, 66.9, 58.9, 16.0; ESI-MS: 303 [Na adduct]; HRMS-ESI: (Na adduct) 303.1573 (actual), 303.1572 (calcd).

(3*S***,4***S***)-4-Benzyloxy-3-(2-methoxyethoxymethoxy)pentan-1-ol, C16H26O5, Bn-OH-33**

Procedure similar to that for **37**, providing the alcohol **Bn-OH-33** in 82% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 7.26–7.34 (5 H, m), 4.85 (1 H, d, *J* 6.9), 4.76 (1 H, d, *J* 6.9), 4.62 (1 H, d, *J* 11.8), 4.51 (1 H, d, *J* 11.8), 3.52–3.89 (8 H, m), 3.38 (3 H, s), 2.74–2.84 (1 H, br s), 1.87 (1 H, ddd, *J* 4.6, 9.1 and 13.9), 1.63 (1 H, ddd, *J* 4.6, 8.8 and 14.1), 1.18 (3 H, d, *J* 6.3); δ_c (75.5 MHz; CDCl₃; Me4Si) 138.5, 128.4, 127.7, 127.6, 96.0, 77.7, 76.5, 71.7, 71.3, 67.4, 59.1 (2 carbons), 32.9, 15.0; ESI-MS: 321 [Na adduct].

(3*S***,4***S***)-4-Benzyloxy-3-(2-methoxyethoxymethoxy)pentanal,** $C_{16}H_{24}O_5$, 7

Procedure similar to that for **keto-10**, providing the pure aldehyde **7** in 97% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 9.72– 9.75 (1 H, m), 7.26–7.32 (5 H, m), 4.75 (2 H, s), 4.57 (1 H, d, *J* 12.1), 4.44 (1 H, d, *J* 11.8), 4.18–4.25 (1 H, m), 3.46–3.72 (5 H, m), 3.33 (3 H, s), 2.52–2.75 (2 H, m), 1.15 (3 H, d, *J* 6.3); δ _C(75.5 MHz; CDCl₃; Me₄Si) 201.1, 138.3, 128.4, 127.8, 127.7, 95.7, 75.0, 74.7, 71.6, 71.1, 67.3, 59.0, 44.5, 14.5.

(2*S***,3***S***,5***Z***,10***S***)-2-benzyloxy-11-***tert***-butyldimethylsilyloxy-3-** (2-methoxyethoxymethoxy)undec-5-ene, C₂₉H₅₂O₅Si, 48

Procedure similar to that for **42**, providing the olefin **48** in 58% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 7.25–7.33 (5 H, m), 5.42 (2) H, d, *J* 5.3), 4.79 (2 H, s), 4.61 (1 H, d, *J* 12.0), 4.50 (1 H, d, *J* 11.9), 3.41–3.71 (8 H, m), 3.37 (3 H, s), 2.25–2.43 (2 H, m), 2.02–2.03 (2 H, m), 1.26–1.57 (4 H, m), 1.19 (3 H, d, *J* 5.3),

1.02–1.10 (1 H, m), 0.88 (9 H, s), 0.86 (3 H, d, *J* 7.2), 0.04 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 138.8, 131.9, 128.3, 127.6, 127.4, 125.8, 95.5, 79.9, 75.8, 71.8, 71.3, 68.4, 67.0, 59.0, 35.7, 32.9, 28.1, 27.8, 27.1, 26.0, 18.4, 16.8, 15.3, −5.3.

(2*S***,3***S***,5***Z***,10***S***)-11-***tert***-Butyldimethylsilyloxy-3-(2** methoxyethoxymethoxy)undec-5-en-2-ol, C₂₂H₄₆O₅Si, OH-48

Procedure similar to that for **OH-TBS-23h**, providing the alcohol **OH-48** in 85% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 5.32– 5.48 (2 H, m), 4.76 (2 H, s), 3.63–3.79 (3 H, m), 3.50–3.53 (2 H, m), 3.27–3.42 (6 H, m), 2.85–2.89 (1 H, br s), 2.12–2.36 (2 H, m), 1.89–2.03 (2 H, m), 1.50–1.54 (1 H, m), 1.22–1.36 (3 H, m), 1.12 (3 H, d, *J* 6.3), 0.98–1.10 (1 H, m), 0.84 (9 H, s), 0.79 (3 H, d, *J* 7.2), 0.01 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 132.3, 124.6, 95.8, 83.9, 71.7, 68.9, 68.3, 67.6, 59.0, 35.6, 32.8, 29.0, 27.7, 26.9, 25.9, 19.1, 18.3, 16.7, −5.4.

(3*S***,5***Z***,10***S***)-11-***tert***-Butyldimethylsilyloxy-3-(2** methoxyethoxymethoxy)undec-5-en-2-one, C₂₂H₄₄O₅Si, 28

Procedure similar to that for **keto-10**, providing the pure ketone **28** in 95% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.28–5.52 (2 H, m), 4.74 (1 H, d, *J* 7.0), 4.69 (1 H, d, *J* 7.0), 4.06 (1 H, dd, *J* 6.1 and 6.3), 3.66–3.70 (2 H, m), 3.47–3.50 (2 H, m), 3.28–3.42 (5 H, m), 2.42 (2 H, dd, *J* 5.9 and 6.6), 2.15 (3 H, s), 1.94–2.03 (2 H, m), 1.50–1.53 (1 H, m), 1.22–1.38 (3 H, m), 1.00–1.03 (1 H, m), 0.86 (9 H, s), 0.82 (3 H, d, *J* 6.7), 0.01 (6 H, s); δ_c (75.5 MHz; CDCl3; Me4Si) 209.2, 133.3, 123.3, 95.2, 82.1, 71.6, 68.3, 67.5, 59.0, 35.7, 32.8, 29.9, 27.7, 26.9, 26.6, 26.0, 18.3, 16.7, −5.4.

4-((1*E***,3***S***,5***Z***10***S***)-11-***tert***-Butyldimethylsilyloxy-3-(2 methoxyethoxymethoxy)-2,10-dimethylundeca-1,5-dienyl)-2 methylthiazole, C27H49O4SNSi, 49**

Procedure similar to that for **39**, providing the coupled olefin **49** in 98% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 6.93 (1 H, s), 6.48 (1 H, s), 5.32–5.48 (2 H, m), 4.72 (1 H, d, *J* 6.9), 4.64 (1 H, d, *J* 6.8), 4.13 (1 H, dd, *J* 6.6 and 6.9), 3.80–3.86 (1 H, m), 3.53–3.64 (3 H, m), 3.30–3.45 (5 H, m), 2.70 (3 H, s), 2.30–2.45 (2 H, m), 1.91–2.05 (5 H, m), 1.22–1.58 (4 H, m), 0.99–1.09 (1 H, m), 0.88 (9 H, m), 0.84 (3 H, d, *J* 6.8), 0.03 (6 H, s); δ _C(75.5 MHz; CDCl₃; Me4Si) 164.6, 152.8, 138.4, 132.0, 125.2, 121.5, 115.9, 92.8, 81.7, 71.8, 68.4, 67.0, 59.0, 35.7, 32.9, 31.9, 27.8, 27.1, 26.0, 19.3, 18.4, 16.8, 13.8, −5.3.

4-((1*E***,3***S***,5***Z***,10***S***)-11-Hydroxy-3-(2-methoxyethoxymethoxy)-** 2,10-dimethylundeca-1,5-dienyl)-2-methylthiazole, C₂₁H₃₅O₄SN, **OH-49**

The silyl ether **49** (2.0 g, 3.9 mmol) was dissolved in 8.0 mL of THF, and 2.0 mL of AcOH and 2.0 mL of H_2O were added to the reaction mixture. After the completion of the reaction as indicated by TLC, the reaction mixture was concentrated *in vacuo* and worked up with ether and water. The combined organic layers were dried $(MgSO₄)$, and concentrated under reduced pressure. The crude hydroxyketone was purified by column chromatography (silica gel, hexanes–ethyl acetate, 4 : 1) to obtain 1.2 g (80%) of the pure alcohol **OH-49**. $\delta_H(300 \text{ MHz};$ CDCl3; Me4Si) 6.95 (1 H, s), 6.49 (1 H, s), 5.34–5.47 (2 H, m), 4.72 (1 H, d, *J* 6.9), 4.64 (1 H, d, *J* 6.8), 4.14 (1 H, t, *J* 6.9), 3.80–3.87 (1 H, m), 3.39–3.64 (5 H, m), 3.38 (3 H, s), 2.71 (3 H, s), 2.31–2.47 (2 H, m), 1.98–2.08 (6 H, m), 1.08–1.60 (5 H, m), 0.89 (3 H, d, *J* 6.7); δ _C(75.5 MHz; CDCl₃; Me₄Si) 164.7, 152.7, 138.5, 131.9, 125.5, 121.5, 115.9, 92.8, 81.8, 71.8, 68.2, 67.0, 59.1, 35.7, 32.8, 31.9, 27.7, 19.2, 16.6, 13.8.

4-((1*E***,3***S***,5***Z***,10***S***)-3-(2-Methoxyethoxymethoxy)-2,10-dimethyl-**11-oxoundeca-1,5-dienyl)-2-methylthiazole, $C_{21}H_{33}O_4 SN$, 3

Procedure similar to that for **keto-10**, providing the pure aldehyde **3** (the $C_7 - C_{21}$ subunit of epothilone) in 98% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 9.58 (1 H, d, *J* 2.0), 6.94 (1 H, s), 6.47 (1 H, s), 5.39–5.43 (2 H, m), 4.71 (1 H, d, *J* 6.8), 4.63 (1 H, d, *J* 6.9), 4.12 (1 H, t, *J* 6.8), 3.78–3.85 (1 H, m), 3.51–3.63 (3 H, m), 3.37 (3 H, s), 2.69 (3 H, s), 2.27–2.42 (3 H, m), 1.97–2.07 (5 H, m), 1.33–1.41 (4 H, m), 1.06 (3 H, d, J 7.0); δ_c (75.5 MHz; CDCl₃; Me4Si) 205.1, 164.6, 152.8, 138.3, 131.1, 126.0, 121.5, 116.0, 92.8, 81.6, 71.8, 67.1, 59.1, 46.2, 32.0, 30.1, 27.4, 26.9, 19.3, 13.8, 13.3; EI-MS: *m/z* 256, 89 [100%, CH₃OCH₂CH₂OCH₂+]; CI-MS: *m/z* 396 [(M + H)⁺], 290 [100%, M + H – HOCH₂OCH₂CH₂OCH₃], 256, 168, 89; HRMS-CI: (M + H) 396.2205 (actual), 396.2209 (calcd).

(2*R***,3***S***)-1-***tert***-Butyldimethylsilyloxy-3-methylpent-4-en-2-ol, C12H26O2Si, 51**

Potassium *tert*-butoxide (56.7 mL, 1.0 M solution, 56.7 mmol) was dissolved in 100.0 mL THF at −78 *◦*C and *trans*-2-butene (12.0 mL, 128.9 mmol) was added to it. *n*-Butyllithium (22.7 mL, 2.5 M solution, 56.7 mmol) was added to it and stirred for 20 min at −45 *◦*C. The reaction mixture was cooled to −78 *◦*C and (+)- *B*-methoxydiisopinocampheylborane $[(+)$ -Ipc₂BOMe] (22.0 g, 69.6 mmol) dissolved in 50.0 mL THF was added to it and stirred for 1 h. Aldehyde **50** (9.0 g, 51.6 mmol) was dissolved in 20.0 mL of THF pre-cooled to −78 *◦*C, transferred to the reaction mixture *via* a cannula at −78 *◦*C and stirred for 3 h. The reaction mixture was oxidized with 27.8 mL 3 M NaOH and 27.7 mL 30% H_2O_2 and stirred overnight. The reaction mixture was worked up with ether and water. The organic layer was dried over MgSO4, concentrated under vacuum and purified by column chromatography (silica gel, hexane–ethyl ether, 3 : 2) to obtain 9.7 g (82%) of pure alcohol 51. $\delta_H(300 \text{ MHz}; \text{CDCl}_3;$ Me4Si) 5.66–5.78 (1 H, m), 4.98–5.07 (2 H, m), 3.61–3.68 (1 H, m), 3.40–3.49 (2 H, m), 2.48 (1 H, d, *J* 2.8), 2.22–2.34 (1 H, m), 1.08 (3 H, d, *J* 6.8), 0.89 (9 H, s), 0.06 (6 H, s); δ_c (75.5 MHz; CDCl3; Me4Si) 140.5, 114.9, 74.7, 64.2, 41.0, 25.9, 18.3, 16.0, −5.4.

(2*R***,3***S***)-1,2-Bis(***tert***-butyldimethylsilyloxy)-3-methylpent-4-ene,** $C_{18}H_{40}O_2Si_2$, TBS-51

Procedure similar to that for **19**, providing the silyl ether **TBS-51** in 89% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.79–5.91 (1 H, m), 4.96–5.09 (2 H, m), 3.44–3.62 (2 H, m), 2.39–2.45 (1 H, m), 0.98 (3 H, d, *J* 6.9), 0.90 (9 H, s), 0.89 (9 H, s), 0.06 (12 H, s).; δ _C(75.5 MHz; CDCl₃; Me₄Si) 142.3, 113.7, 76.5, 75.4, 40.3, 26.0, 25.9, 18.4, 18.2, 13.6, −4.7 (2 carbons), −5.3, −5.4.

(2*R***,3***R***)-3,4-Bis(***tert***-butyldimethylsilyloxy)-2-methylbutanal,** $C_{17}H_{38}O_3Si_2$, 52

NMO (6.8 g, 58.2 mmol) was added to the olefinic ether **TBS-51** (10.0 g, 29.1 mmol) dissolved in acetone–water (4 : 1, 60.0 mL) at 0 *◦*C. OsO4 (0.7 g, 2.9 mmol) was added to the above solution and stirred for 6 h at room temperature, and the product was extracted with ether, washed with a saturated solution of sodium sulfite and dried over MgSO4. The solvent was removed under aspirator vacuum. The crude product was dissolved in acetone–water (4 : 1, 50.0 mL) and stirred at room temperature, followed by the slow addition of $NaIO₄$ (12.4 g, 58.2 mmol). The mixture was stirred for 0.5 h at room temperature and the product was extracted with ether, and washed with a saturated solution of sodium thiosulfate. The solvent was removed under aspirator vacuum and the crude product was purified by column chromatography (silica gel, hexane–ethyl acetate, 3 : 2) to obtain 9.0 g (90%) of **52.** $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 9.77 (1 H, s), 4.12–4.18 (1 H, m), 3.62 (1 H, dd, *J* 4.8 and 10.0), 3.50 (1 H, dd, *J* 6.0 and 9.0), 2.54–2.62 (1 H, m), 1.06 (3 H, d, *J* 6.9), 0.87 (9 H, s), 0.86 (9 H, s), 0.05 (12 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 204.6, 72.2, 64.4, 49.6, 25.9, 25.7, 18.3, 18.0, 7.6, −4.2, −4.9, −5.4 (2 carbons).

(2*R***,3***S***,4***Z***)-7-Benzyloxy-1,2-bis(***tert***-butyldimethylsilyloxy)-3** methylhept-4-ene, C₂₇H₅₀O₃Si₂, 53

A mixture of 1-benzyloxy-3-iodopropane (2.0 g, 7.4 mmol) and Ph₃P (1.9 g, 7.4 mmol) was heated in xylene at 100 $\rm{°C}$ for 10 h. Evaporation of the xylene provided white crystals of the phosphonium salt. The solid was submerged in THF with occasional heating. The solution was cooled to 0 *◦*C and NaHMDS (6.9 mL, 6.9 mmol) was added slowly. The clear red solution was stirred for 20 min at room temperature and recooled to 0 *◦*C. Aldehyde **52** (1.7 g, 4.9 mmol) was added to the solution and stirred at 0 *◦*C for 2 h. The reaction mixture was extracted with ether and water. The combined organic layers were dried, concentrated under vacuum and purified by column chromatography (silica gel, hexanes–ethyl acetate, 9 : 1) to provide 1.8 g (77%) of olefin 53. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 7.26–7.38 (5 H, m), 5.33–5.45 (2 H, m), 4.51 (2 H, s), 3.41–3.56 (5 H, m), 2.64–2.76 (1 H, m), 2.32–2.46 (2 H, m), 0.93 (3 H, d, *J* 6.8), 0.89 (18 H, s), 0.05 (12 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 138.6, 129.5, 128.4, 127.6, 127.5, 124.7, 76.9, 72.9, 70.1, 65.7, 34.2, 28.2, 26.1, 26.0, 18.4, 18.3, 15.0, −4.0, −4.7, −5.2, −5.3.

(5*S***,6***R***)-6,7-Bis(***tert***-butyldimethylsilyloxy)-5-methylheptan-1 ol, C20H46O3Si2, OH-6**

Benzyl ether **53** (1.6 g, 3.3 mmol) was dissolved in ethyl acetate (10.0 mL). 0.15 g of 10% w/v palladium-on-charcoal was added to it and hydrogen gas bubbled through the reaction mixture for 6 h. The reaction mixture was filtered over silica gel and purified by column chromatography (silica gel, hexanes–ethyl acetate, 3 : 1) to obtain 0.78 g (60%) of pure alcohol **OH-6**. $\delta_{\rm H}$ (300 MHz; CDCl3; Me4Si) 3.65 (2 H, t, *J* 6.6), 3.57–3.61 (1 H, m), 3.47 (2 H, d, *J* 6.4), 1.51–1.67 (3 H, m), 1.30–1.45 (2 H, m), 1.16–1.29 (2 H, m), 0.88 (9 H, s), 0.87 (9 H, s), 0.82 (3 H, d, *J* 6.8), 0.04 (12 H, s); δ_c (75.5 MHz; CDCl₃; Me₄Si) 75.9, 65.3, 63.1, 35.3, 33.4, 33.1, 26.0, 25.9, 23.8, 18.2 (2 carbons), 13.3, −3.9, −4.8, −5.2, −5.4.

(2*R***,3***S***)-1,2-Bis(***tert***-butyldimethylsilyloxy)-7-iodo-3-methylheptane,** $C_{20}H_{45}IO_2Si_2$ **, 6**

Alcohol $OH-6$ (0.3 g, 0.8 mmol) was dissolved in CH₂Cl₂. Iodine (0.3 g, 1.15 mmol), triphenylphosphine (0.3 g, 1.2 mmol) and imidazole (0.2 g, 2.3 mmol) were added and stirred for 8 h. After the completion of the reaction as indicated by TLC, the reaction mixture was worked up with ether and water. The organic layers were dried, concentrated under vacuum and purified by column chromatography (silica gel, hexanes–ethyl acetate, 19 : 1) to afford 0.3 g (78%) of 6. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{Me}_4\text{Si})$ 3.57–3.62 (1 H, m), 3.47 (2 H, d, *J* 6.1), 3.47 (2 H, t, *J* 7.0), 1.77–1.86 (2 H, m), 1.60–1.70 (1 H, m), 1.34–1.47 (3 H, m), 1.15–1.19 (1 H, m), 0.89 (9 H, s), 0.88 (9 H, s), 0.82 (3 H, d, *J* 6.6), 0.05 (12 H, s); δ _C(75.5 MHz; CDCl₃; Me₄Si) 75.8, 65.2, 35.1, 33.8, 32.5, 28.6, 26.0, 25.9, 18.4 (2 carbons), 13.3, 7.2, −3.9, −4.8, −5.2, −5.3.

(2*S***,3***S***,5***Z***,10***S***,11***R***)-2-Benzyloxy-11,12-bis(***tert***butyldimethylsilyloxy)-3-(2-methoxyethoxymethoxy)-10** methyldodec-5-ene, $C_{36}H_{68}O_6Si_2$, 54

Procedure similar to that for **42**, providing the olefin **54** in 85% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 7.26–7.36 (5 H, m), 5.43 (2 H, ddd, *J* 6.4, 11.4 and 17.4), 4.79 (2 H, s), 4.61 (1 H, d, *J* 11.9), 4.50 (1 H, d, *J* 11.8), 3.47–3.53 (4 H, m), 3.57–3.73 (5 H, m), 3.37 (3 H, s), 2.23–2.44 (2 H, m), 1.96–2.08 (2 H, m), 1.62–1.65 (1 H, m), 1.27–1.43 (3 H, m), 1.09–1.42 (1 H, m), 1.19 (3 H, d, *J* 5.5), 0.90 (9 H, s), 0.89 (9 H, s), 0.81 (3 H, d, *J* 6.6), 0.05 (6 H, s), 0.04 (6 H, s); $δ_C(75.5 MHz; CDCl₃; Me₄Si) 138.8, 132.0,$ 128.3, 127.6, 127.4, 125.7, 95.5, 79.9, 75.9, 71.8, 71.3, 67.0, 65.4, 59.0 (2 carbons), 35.3, 33.5, 28.1, 27.7 (2 carbons), 26.0, 25.9, 18.4, 18.2, 15.3, 13.4, −4.0, −4.7, −5.2, −5.3. ESI: *m*/*z* 675 [Na adduct]; HRMS-ESI: (Na adduct) 675.4460 (actual), 675.4452 (calcd).

(2*S***,3***S***,5***Z***,10***S***,11***R***)-11,12-Bis(***tert***-butyldimethylsilyloxy)-3- (2-methoxyethoxymethoxy)-10-methyldodec-5-en-2-ol, C29H62O6Si2, OH-54**

Procedure similar to that for **OH-TBS-23h**, providing **OH-54** in 95% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.35–5.51 (2 H, m), 4.79 (2 H, s), 3.67–3.82 (3 H, m), 3.52–3.60 (3 H, m), 3.44–3.46 (2 H, m), 3.37 (3 H, s), 3.35–3.36 (1 H, m), 2.95–2.97 (1 H, br s), 2.18–2.38 (2 H, m), 1.98–2.05 (2 H, m), 1.58–1.66 (1 H, m), 1.16–1.41 (4 H, m), 1.14 (3 H, d, *J* 6.4), 0.87 (9 H, s), 0.86 (9 H, s), 0.79 (3 H, d, *J* 6.8), 0.03 (3 H, s), 0.02 (3 H, s), 0.01 $(6 H, s)$; δ_c (75.5 MHz; CDCl₃; Me₄Si) 132.4, 124.5, 95.8, 84.0, 75.8, 71.7, 68.9, 67.6, 59.0, 35.2, 33.4, 29.1, 27.7, 27.6, 26.0, 25.9, 19.1, 18.3, 18.2, 13.3, −4.0, −4.8, −5.3, −5.4; ESI: *m*/*z* 585 [Na adduct]; HRMS-ESI: (Na adduct) 585.3981 (actual), 585.3983 (calcd).

(3*S***,5***Z***,10***S***,11***R***)-11,12-Bis(***tert***-butyldimethylsilyloxy)-3-(2 methoxyethoxymethoxy)-10-methyldodec-5-en-2-one,** $C_{29}H_{60}O_6Si_2$, 4

Procedure similar to that for **keto-10**, providing the pure aldehyde **4** in 98% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 5.48– 5.56 (1 H, m), 5.32–5.40 (2 H, m), 4.77 (1 H, d, *J* 7.1), 4.73 (1 H, d, *J* 7.1), 4.09 (1 H, t, *J* 6.2), 3.45–3.73 (6 H, m), 3.37 (3 H, s), 2.45 (2 H, t, *J* 6.6), 2.17 (3 H, s), 1.99–2.05 (2 H, m), 1.61–1.64 (1 H, m), 1.12–1.42 (4 H, m), 0.88 (9 H, s), 0.87 (9 H, s), 0.80 (3 H, d, *J* 6.8), 0.07 (3 H, s), 0.04 (6 H, s), 0.03 $(3 H, s); \delta_c(75.5 MHz; CDCl₃; Me₄Si) 209.2, 133.3, 123.2, 95.2,$ 82.1, 75.8, 71.7, 67.5, 65.3, 59.0, 35.2, 33.4, 29.9, 27.6, 27.5, 26.5, 26.0, 25.9, 18.3, 18.2, 13.3, −4.0, −4.8, −5.3, −5.4; EI-MS: *m*/*z* 427, 147, 133, 89 [100%, CH₂OCH₂CH₂OCH₃+], 73, 59; CI-MS: m/z 561 [(M + H)⁺], 485 [(M + H – HOCH₂CH₂OCH₃)⁺], 353 [(M + H – HOCH₂CH₂OCH₃ – HOSi(CH₃)₂C(CH₃)₃)⁺], 221, 89 [100%, $CH_2OCH_2CH_2OCH_3^+$]; HRMS-CI: (M + H) 561.4008 (actual), 561.4007 (calcd).

4-(((1*E***,3***S***,5***Z***,10***S***,11***R***)-11,12-Bis(***tert***-butyldimethylsilyloxy)- 3-(2-methoxyethoxymethoxy)-2,10-dimethyldodeca-1,5-dienyl)-** 2-methylthiazole, C₃₄H₆₅O₅SNSi₂, 55

Procedure similar to that for **39**, providing the coupled olefin **55** in 83% yield. $\delta_H(300 \text{ MHz}; \text{CDCl}_3; \text{ Me}_4\text{Si})$ 6.92 (1 H, s), 6.48 (1 H, s), 5.40 (2 H, ddd, *J* 6.9, 12.2 and 18.7), 4.71 (1 H, d. *J* 6.9), 4.63 (1 H, d, *J* 6.9), 4.12 (1 H, t, *J* 6.8), 3.78–3.86 (1 H, m), 3.52–3.63 (4 H, m), 3.45–3.47 (2 H, m), 3.37 (3 H, s), 2.69 (3 H, s), 2.30–2.46 (2 H, m), 1.94–2.05 (5 H, m), 1.58–1.63 (1 H, m), 1.07–1.43 (4 H, m), 0.87 (9 H, s), 0.86 (9 H, s), 0.79 (3 H, d, *J* 6.9), 0.03 (6 H, s), 0.02 (6 H, s); δ_c (75.5 MHz; CDCl₃; Me4Si) 164.5, 152.8, 138.4, 132.0, 125.2, 121.5, 115.9, 92.7, 81.7, 75.8, 71.8, 67.0, 65.3, 59.0, 35.2, 33.4, 31.9, 27.8, 27.7, 26.0, 25.9, 19.2, 18.4, 18.2, 13.8, 13.4, −4.0, −4.8, −5.3, −5.4; ESI: *m*/*z* 656, 678 [Na adduct]; HRMS-ESI: (Na adduct) 656.4202 (actual), 656.4200 (calcd).

4-((1*E***,3***S***,5***Z***,10***S***)-3-(2-Methoxyethoxymethoxy)-2,10 dimethyl-11-oxoundeca-1,5-dienyl)-2-methylthiazole,** $C_{21}H_{33}O_4$ NS, 3

The silyl ether **55** (0.5 g, 0.7 mmol) was dissolved in 2.0 mL of THF, and 0.5 mL of AcOH and 0.5 mL of $H₂O$ were added to the reaction mixture. After the completion of the reaction as indicated by TLC, the reaction mixture was concentrated *in vacuo* and worked up with ether and water. The combined organic layers were dried $(MgSO₄)$, and concentrated under reduced pressure. The crude diol **OH-55** was utilized for the next step without further purification. The diol **OH-55** was dissolved in benzene (4.0 mL) and stirred at 25 *◦*C. Pb(OAc)4 (0.7 g, 1.5 mmol) was added to the reaction mixture and stirred for 20 min. After the completion of the reaction as indicated by TLC, the reaction mixture was worked up with ether and

water. The combined organic layers were dried over MgSO₄ and concentrated under vacuum. The crude product was purified by column chromatography (silica gel, hexanes–ethyl acetate, 4 : 1), to provide the required $C_7 - C_{21}$ subunit of epothilone 3 in 0.2 g (70% overall yield for the two steps). $\delta_H(300 \text{ MHz}; \text{CDCl}_3;$ Me4Si) 9.58 (1 H, d, *J* 2.0), 6.94 (1 H, s), 6.47 (1 H, s), 5.39–5.43 (2 H, m), 4.71 (1 H, d, *J* 6.8), 4.63 (1 H, d, *J* 6.9), 4.12 (1 H, t, *J* 6.8), 3.78–3.85 (1 H, m), 3.51–3.63 (3 H, m), 3.37 (3 H, m), 2.69 (3 H, s), 2.27–2.42 (3 H, m), 1.97–2.07 (5 H, m), 1.33–1.41 (4 H, m), 1.06 (3 H, d, *J* 7.0); δ _C(75.5 MHz; CDCl₃; Me₄Si) 205.1, 164.6, 152.8, 138.3, 131.1, 126.0, 121.5, 116.0, 92.8, 81.6, 71.8, 67.1, 59.1, 46.2, 32.0, 30.1, 27.4, 26.9, 19.3, 13.8, 13.3; EI-MS: *m*/*z* 256, 89 [100%, CH₃OCH₂CH₂OCH₂⁺]; CI-MS: *m*/*z* 396 $[(M + H)^+]$, 290 [100%, M + H – HOCH₂OCH₂CH₂OCH₃], 256, 168, 89; HRMS-CI: (M + H) 396.2205 (actual), 396.2209 (calcd).

Acknowledgements

This paper is dedicated to the memory of Professor Ian P. Rothwell. Financial assistance from the Herbert C. Brown Center for Borane Research**²⁸** and the Aldrich Chemical Company are gratefully acknowledged.

References and notes

- 1 D. M. Bollag, P. A. McQueney, J. Zhu, O. Hensens, L. Koupal, J. Liesch, M. Goetz, E. Lazarides and C. M. Woods, *Cancer Res.*, 1995, **55**, 2325.
- 2 For reviews on epothilones, see: (*a*) R. Altaha, T. Fojo, E. Reed and J. Abraham, *Curr. Pharm. Des.*, 2002, **8**, 1707; (*b*) S. J. Stachel, K. Biswas and S. J. Danishefsky, *Curr.. Pharm. Des.*, 2001, **7**, 1277; (*c*) J. Mulzer, *Monatsh. Chem.*, 2000, **131**, 205; (*d*) K. C. Nicolaou, F. Roschanger and D. Vourloumis, *Angew. Chem., Intl. Ed.*, 1998, **37**, 2015; (*e*) M. R. V. Finlay, *Chem. Ind.*, 1997, 991; (*f*) L. Wessjohann, *Angew. Chem., Intl. Ed. Engl.*, 1997, **36**, 715, and references therein. For total syntheses, see:; (*g*) A. Balog, D. F. Meng, T. Kamenecka, P. Bertinato, D. S. Su, E. J. Sorensen and S. J. Danishefsky, *Angew. Chem., Int. Ed. Engl.*, 1996, **35**, 2801; (*h*) K. C. Nicolaou, N. Winssinger, J. Pastor, S. Ninkovic, F. Sarabia, Y. He, D. Vourloumis, Z. Yang, T. Li, P. Giannakakou and E. Hamel, *Nature*, 1997, **387**, 268; (*i*) D. Schinzer, A. Limberg, A. Bauer, O. M. Bohm and M. Cordes, *Angew. Chem., Intl. Ed. Engl.*, 1997, **36**, 523; (*j*) K. C. Nicolaou, Y. He, D. Vourloumis, H. Vallberg, F. Roschangar, F. Sarabia, S. Ninkovic, Z. Yang and J. I. Trujillo, *J. Am. Chem. Soc.*, 1997, **119**, 7960; (*k*) K. C. Nicolaou, S. Ninkovic, F. Sarabia, D. Vourloumis, Y. He, H. Vallberg, M. R. V. Finlay and Z. Yang, *J. Am. Chem. Soc.*, 1997, **119**, 7974; (*l*) D. F. Meng, P. Bertinato, A. Balog, D. S. Su, T. Kamenecka, E. J. Sorensen and S. J. Danishefsky, *J. Am. Chem. Soc.*, 1997, **119**, 10073; (*m*) S. A. May and P. A. Grieco, *Chem. Commun.*, 1998, 1597; (*n*) D. Schinzer, A. Bauer, O. M. Bohm, A. Limberg and M. Cordes, *Chem. Eur. J.*, 1999, **5**, 2483; (*o*) J. Mulzer, G. Karig and P. Pojarliev, *Tetrahedron Lett.*, 2000, **41**, 7635; (*p*) B. Zhu and J. S. Panek, *Org. Lett.*, 2000, **2**, 2575; (*q*) D. Sawada, M. Kanai and M. Shibasaki, *J. Am. Chem. Soc.*, 2000, **122**, 10521; (*r*) A. Furstner, C. Mathes and C. W. Lehmann, *Chem. Eur. J.*, 2001, **7**, 5299; (*s*) M. Valluri, R. M. Hindupur, P. Bijoy, G. Labadie, J. C. Jung and M. A. Avery, *Org. Lett.*, 2001, **3**, 3607; (*t*) R. M. Hindupur, B. Panicker, M. Valluri and M. A. Avery, *Tetrahedron Lett.*, 2001, **42**, 7341; (*u*) J. W. Bode and E. M. Carreira, *J. Am. Chem. Soc.*, 2001, **123**, 3611; (*v*) Z. Y. Liu, Z. C. Chen, C. Z. Yu, R. F. Wang, R. Z. Zhang, C. S. Huang, Z. Yan, D. R. Cao, J. B. Sun and G. Li, *Chem. Eur. J.*, 2002, **8**, 3747.
- 3 For recent analog syntheses, see: (*a*) D. Schinzer, E. Bourguet and S. Ducki, *Chem. Eur. J.*, 2004, **10**, 3217; (*b*) D. Schinzer, O. M. Bohm, K. H. Altmann andM.Wartmann, *Synlett*, 2004, 375; (*c*) G. Koch, O. Loiseleur and K. H. Altmann, *Synlett*, 2004, 693; (*d*) S. D. Dong, K. Sundermann, K. M. J. Smith, J. Petryka, F. H. Liu and D. C. Myles, *Tetrahedron Lett.*, 2004, **45**, 1945; (*e*) T. Ganesh, J. K. Schilling, R. K. Palakodety, R. Ravindra, N. Shanker, S. Bane and D. G. I. Kingston, *Tetrahedron*, 2003, **59**, 9979; (*f*) D. Quintard, P. Bertrand, S. Vielle, E. Raimbaud, P. Renard, B. Pfeiffer and J. P. Gesson, *Synlett*, 2003, 2033; (*g*) T. C. Chou, H. J. Dong, A. Rivkin, F. Yoshimura, A. E. Gabarda, Y. S. Cho, W. P. Tong and S. J. Danishefsky, *Angew. Chem., Int. Ed.*, 2003, **42**, 4761; (*h*) K. C. Nicolaou, P. K. Sasmal, G. Rassias, M. V. Reddy, K. H. Altmann, M. Wartmann, A. O'Brate and P. Giannakakou, *Angew. Chem., Int. Ed.*, 2003, **42**, 3515; (*i*) C. N. Boddy, T. L. Schneider, K. Hotta, C. T. Walsh and C. Khosla, *J. Am.*

Chem. Soc., 2003, **125**, 3428; (*j*) U. Karama and G. Hofle, *Eur. J. Org. Chem.*, 2003, **6**, 1042; (*k*) L. Tang, R. G. Qiu, Y. Li and L. Katz, *J. Antibiot.*, 2003, **56**, 16; (*l*) R. E. Taylor, Y. Chen, A. Beatty, D. C. Myles and Y. Q. Zhou, *J. Am. Chem. Soc.*, 2003, **125**, 26.

- 4 (*a*) P. V. Ramachandran, M. V. R. Reddy and H. C. Brown, *Pure Appl. Chem.*, 2003, **75**, 1263; (*b*) P. V. Ramachandran, *Aldrichimica Acta*, 2002, **35**, 23; (*c*) H. C. Brown and P. V. Ramachandran, *J. Organomet. Chem.*, 1995, **500**, 1.
- 5 (*a*) P. V. Ramachandran, B. Prabhudas, J. S. Chandra and M. V. R. Reddy, *J. Org. Chem.*, 2004, **69**, 6294; (*b*) P. V. Ramachandran, J. S. Chandra and M. V. R. Reddy, *J. Org. Chem.*, 2002, **67**, 7547; (*c*) M. V. R. Reddy, J. P. Rearick, N. Hoch and P. V. Ramachandran, *Org. Lett.*, 2001, **3**, 19; (*d*) M. V. R. Reddy, A. J. Yucel and P. V. Ramachandran, *J. Org. Chem.*, 2001, **66**, 2512; (*e*) P. V. Ramachadran, M. V. R. Reddy and H. C. Brown, *Tetrahedron Lett.*, 2000, **41**, 583.
- 6 (*a*) P. V. Ramachandran, B. Prabhudas, J. S. Chandra, M. V. R. Reddy and H. C. Brown, *Tetrahedron Lett.*, 2004, **45**, 1011; (*b*) P. V. Ramachandran, B. Prabhudas, D. Pratihar, J. S. Chandra and M. V. R. Reddy, *Tetrahedron Lett.*, 2003, **44**, 3745.
- 7 (*a*) P. V. Ramachandran and H. C. Brown, *ACS Symp. Ser.*, 1996, **641**, 84; (*b*) H. C. Brown and P. V. Ramachandran, *Acc. Chem. Res.*, 1992, **25**, 16.
- 8 P. V. Ramachandran, A. V. Teodorovic, M. V. Rangaishenvi and H. C. Brown, *J. Org. Chem.*, 1992, **57**, 2379.
- 9 P. V. Ramachandran, B. Q. Gong, A. V. Teodorovic and H. C. Brown, *Tetrahedron: Asymmetry*, 1994, **5**, 1061.
- 10 P. V. Ramachandran, Z. H. Lu and H. C. Brown, *Tetrahedron Lett.*, 1997, **38**, 761.
- 11 (*a*) P. V. Ramachandran, S. Pitre and H. C. Brown, *J. Org. Chem.*, 2002, **67**, 5315; (*b*) Z. Wang, B. La, J. M. Fortunak, X.-J. Meng and G. W. Kabalka, *Tetrahedron Lett.*, 1998, **39**, 5501.
- 12 D. B. Dess and J. C. Martin, *J. Am. Chem. Soc.*, 1991, **13**, 7277.
- 13 (*a*) R. W. Hoffmann, *Pure Appl. Chem.*, 1988, **60**, 123; (*b*) Y. Yamamoto and N. Asao, *Chem. Rev.*, 1993, **93**, 2207; (*c*)W. R. Roush, in: *Houben-Weyl Methods of Organic Chemistry*, Georg Thieme Verlag, Stuttgart, 1995, vol. E 21, p. 1410.
- 14 (*a*) A. B. Smith, C. M. Adams, S. A. L. Barbosa and A. P. Degnan, *J. Am. Chem. Soc.*, 2003, **125**, 350; (*b*) J. D. White, P. R. Blakemore, N. J. Green, E. B. Hauser, M. A. Holoboski, L. E. Keown, C. S. N. Kolz and B. W. Phillips, *J. Org. Chem.*, 2002, **67**, 7750.
- 15 H. C. Brown and P. K. Jadhav, *Tetrahedron Lett.*, 1984, **25**, 1215.
- 16 W. W. Epstein and C. D. Poulter, *Phytochemistry*, 1973, **12**, 737.
- 17 G. R. Pettit, C. L. Herald, D. L. Doubek, D. L. Herald, E. Arnold and J. Clardy, *J. Am. Chem. Soc.*, 1982, **104**, 6846.
- 18 T. Nakata, S. Nagao and T. Oishi, *Tetrahedron Lett.*, 1985, **26**, 6465. 19 N. B. Perry, J. W. Blunt, M. H. G. Munro and L. K. Pannell, *J. Am. Chem. Soc.*, 1988, **110**, 4851.
- 20 D. Schinzer, A. Limberg and O. M. Bohm, *Chem. Eur. J.*, 1996, **2**, 1477.
- 21 For reviews on cuprate coupling, see:(*a*) M. Tamura and J. Kochi, *Synthesis*, 1971, 303; (*b*) G. Fouquet andM. Schlosser, *Angew. Chem., Int. Ed. Engl.*, 1974, **13**, 82; (*c*) M. Schlosser, *Angew. Chem., Int. Ed. Engl.*, 1974, **13**, 701.
- 22 (*a*) H. C. Brown, P. K. Jadhav and K. S. Bhat, *J. Am Chem. Soc.*, 1988, **110**, 1535; (*b*) A. L. Smith, E. N. Pitsinos, C. K. Hwang, Y. Mizuno, H. Saimoto, G. R. Scarlato, T. Suzuki and K. C. Nicolaou, *J. Am. Chem. Soc.*, 1993, **115**, 7612; (*c*) The reagent used was prepared from $(-)-\alpha$ -pinene.
- 23 Review: J. B. Arterburn, *Tetrahedron*, 2001, **57**, 9765, and references therein.
- 24 The $C_{17}-C_{21}$ subunit 4-chloromethyl-2-methylthiazole **5** was prepared from 1,3-dichloroacetone and thioacetamide using a literature procedure: F. E. Hooper and T. B. Johnson, *J. Am. Chem. Soc.*, 1934, **56**, 470.
- 25 Z. Zhao, G. R. Scarlato and R. W. Armstrong, *Tetrahedron Lett.*, 1991, **32**, 1609.
- 26 (*a*) H. C. Brown and K. S. Bhat, *J. Am Chem. Soc.*, 1986, **108**, 5919; (*b*) The reagent used was prepared from (−)-a-pinene.
- 27 3-Benzyloxypropyl iodide was conveniently prepared in 2 steps starting from 1,3-propanediol, by monobenzylation followed by iodination.
- 28 Contribution #38 from the Herbert C. Brown Center for Borane Research.